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Search stratagies
10/535,603
8/29/2006
CASREACT 7...
no usable hits

closest artols

Welcome to STN International! Enter x:x

LOGINID:SSPTAMPB1626

PASSWORD:

TERMINAL (ENTER 1, 2, 3, OR ?):2

```
Web Page URLs for STN Seminar Schedule - N. America
NEWS
NEWS
                "Ask CAS" for self-help around the clock
                New STN AnaVist pricing effective March 1, 2006
NEWS
        FEB 27
NEWS 4
        APR 04
                STN AnaVist $500 visualization usage credit offered
        MAY 10 CA/CAplus enhanced with 1900-1906 U.S. patent records
NEWS 5
NEWS 6 MAY 11 KOREAPAT updates resume
        MAY 19 Derwent World Patents Index to be reloaded and enhanced
     7
NEWS
               IPC 8 Rolled-up Core codes added to CA/CAplus and
NEWS 8 MAY 30
                USPATFULL/USPAT2
                The F-Term thesaurus is now available in CA/CAplus
NEWS 9 MAY 30
                The first reclassification of IPC codes now complete in
NEWS 10 JUN 02
                INPADOC
        JUN 26
                TULSA/TULSA2 reloaded and enhanced with new search and
NEWS 11
                and display fields
                Price changes in full-text patent databases EPFULL and PCTFULL
NEWS 12
        JUN 28
                CHEMSAFE reloaded and enhanced
NEWS 13
        JUL 11
                FSTA enhanced with Japanese patents
NEWS 14
        JUL 14
                Coverage of Research Disclosure reinstated in DWPI
        JUl 19
NEWS 15
NEWS 16 AUG 09
                INSPEC enhanced with 1898-1968 archive
NEWS 17 AUG 28 ADISCTI Reloaded and Enhanced
             JUNE 30 CURRENT WINDOWS VERSION IS V8.01b, CURRENT
NEWS EXPRESS
             MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP),
             AND CURRENT DISCOVER FILE IS DATED 26 JUNE 2006.
             STN Operating Hours Plus Help Desk Availability
```

NEWS HOURS STN Operating Hours Plus Help Desk Availability
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NEWS IPC8 For general information regarding STN implementation of IPC 8
NEWS X25 X.25 communication option no longer available

Enter NEWS followed by the item number or name to see news on that specific topic.

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FILE 'HOME' ENTERED AT 15:08:49 ON 29 AUG 2006

=> file casreact COST IN U.S. DOLLARS

FULL ESTIMATED COST

SINCE FILE TOTAL
ENTRY SESSION
0.21 0.21

FILE 'CASREACT' ENTERED AT 15:09:02 ON 29 AUG 2006

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FILE CONTENT: 1840 - 27 Aug 2006 VOL 145 ISS 9

New CAS Information Use Policies, enter HELP USAGETERMS for details.

Some CASREACT records are derived from the ZIC/VINITI database (1974-1991) provided by InfoChem, INPI data prior to 1986, and Biotransformations database compiled under the direction of Professor Dr. Klaus Kieslich.

This file contains CAS Registry Numbers for easy and accurate substance identification.

```
chain nodes :
11 12 13
                                                         26
                                                             27
                                                                 28
                                                                     29
                                                                         30
                                                                             31
          14
               15
                  16
                      17
                          18
                              19
                                  20
                                      21
                                          22
                                             23
                                                 24
                                                     25
  33
                  37
                                                     46
                                                         47
                                                             48
                                                                 49
                                                                     50
32
           35
               36
                       38
                          39
                              40
                                  41
                                      42
                                          43
                                              44
                                                  45
       34
59 60 61
           62
               63
                   66
                       67
                          68
                              69
                                  70
ring nodes :
             6 7 8
           5
                      9 10 53 54 55 56 57 58
1 2 3 4
chain bonds :
          7-11 8-12 9-14 10-13 15-16 16-17 17-18 18-19 18-31 19-20
1-15 1-28
20-21 21-22 22-23 22-30 23-24 24-25
                                                                         33-48
                                        25-26 26-27 26-29 32-33 33-34
             36-37
34 - 35
      35-36
                   37-38
                          37-49
                                 38-39
                                        39-40
                                              40-41
                                                     41-42
                                                            41-50
                                                                   42-43
                                                                          43-44
44-45
      45-46
             45-51 46-47 47-52 53-63 54-61 55-60 56-62 57-59 66-69 67-69
68-69
      69-70
```

```
ring bonds :
1-6 1-2 2-3 3-4 4-5 4-10 5-6 5-7 7-8 8-9 9-10 53-54 53-58 54-55 55-56
 56-57 57-58
exact/norm bonds :
1-6 1-2 2-3 3-4 5-6 9-14 47-52 53-63 56-62
exact bonds :
1-15 1-28 7-11 8-12 10-13 15-16 16-17 17-18 18-19 18-31 19-20 20-21
21-22 22-23 22-30 23-24 24-25 25-26 26-27 26-29 32-33 33-34 33-48 34-35
35-36 36-37 37-38 37-49 38-39 39-40 40-41 41-42 41-50 42-43 43-44 44-45
45-46 45-51 46-47 54-61 55-60 57-59 66-69 67-69 68-69 69-70
normalized bonds :
4-5 4-10 5-7 7-8 8-9 9-10 53-54 53-58 54-55 55-56 56-57 57-58
Match level:
1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:Atom 8:Atom 9:Atom 10:Atom
11:CLASS 12:CLASS 13:CLASS 14:CLASS 15:CLASS 16:CLASS 17:CLASS 18:CLASS
19:CLASS 20:CLASS 21:CLASS 22:CLASS 23:CLASS 24:CLASS 25:CLASS 26:CLASS
27:CLASS 28:CLASS 29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS
35:CLASS 36:CLASS 37:CLASS 38:CLASS 39:CLASS 40:CLASS 41:CLASS 42:CLASS
43:CLASS 44:CLASS 45:CLASS 46:CLASS 47:CLASS 48:CLASS 49:CLASS 50:CLASS
51:CLASS 52:CLASS 53:Atom 54:Atom 55:Atom 56:Atom 57:Atom 58:Atom 59:CLASS
60:CLASS 61:CLASS 62:CLASS 63:CLASS 66:CLASS 67:CLASS 68:CLASS 69:CLASS
70:CLASS
fragments assigned product role:
containing 1
fragments assigned reactant/reagent role:
containing 32
containing 53
containing 66
node mappings:
9:56
L1
       STRUCTURE UPLOADED
=> s l1 ful
FULL SEARCH INITIATED 15:10:13 FILE 'CASREACT'
SCREENING COMPLETE - 0 REACTIONS TO VERIFY FROM 0 DOCUMENTS
                                                              0 DOCS
                O VERIFIED O HIT RXNS
100.0% DONE
SEARCH TIME: 00.00.01
            O SEA SSS FUL L1 ( O REACTIONS)
L2
=>
Uploading C:\Program Files\Stnexp\Queries\10535603\10535606narrow1.str
```

```
chain nodes :
11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31
32 33 34 35 36
ring nodes :
1 2 3 4 5 6 7 8 9 10
chain bonds :
1-15 \quad 1-28 \quad 7-11 \quad 8-12 \quad 9-14 \quad 10-13 \quad 15-16 \quad 16-17 \quad 17-18 \quad 18-19 \quad 18-31 \quad 19-20
20-21 21-22 22-23 22-30 23-24 24-25 25-26 26-27 26-29 32-35 33-35 34-35
35-36
ring bonds :
1-6 1-2 2-3 3-4 4-5 4-10 5-6 5-7 7-8 8-9 9-10
exact/norm bonds :
1-6 1-2 2-3 3-4 5-6 9-14
exact bonds :
1-15 1-28 7-11 8-12 10-13 15-16 16-17 17-18 18-19 18-31 19-20 20-21
21-22 22-23 22-30 23-24 24-25 25-26 26-27 26-29 32-35 33-35 34-35 35-36
normalized bonds :
4-5 4-10 5-7 7-8 8-9 9-10
Match level :
1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:Atom 8:Atom 9:Atom 10:Atom
11:CLASS 12:CLASS 13:CLASS 14:CLASS 15:CLASS 16:CLASS 17:CLASS 18:CLASS
                  21:CLASS 22:CLASS 23:CLASS 24:CLASS 25:CLASS 26:CLASS
19:CLASS 20:CLASS
                  29:CLASS 30:CLASS 31:CLASS 32:CLASS 33:CLASS 34:CLASS
27:CLASS 28:CLASS
35:CLASS 36:CLASS
fragments assigned product role:
containing 1
fragments assigned reactant/reagent role:
containing 32
```

L3STRUCTURE UPLOADED

=> s 13 ful

FULL SEARCH INITIATED 15:11:50 FILE 'CASREACT'

O REACTIONS TO VERIFY FROM SCREENING COMPLETE -0 DOCUMENTS

100.0% DONE 0 DOCS 0 VERIFIED 0 HIT RXNS

SEARCH TIME: 00.00.01

L4O SEA SSS FUL L3 (0 REACTIONS)

=> file reg

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION

FULL ESTIMATED COST 223.36 223.57

FILE 'REGISTRY' ENTERED AT 15:14:29 ON 29 AUG 2006 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2006 American Chemical Society (ACS)

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28 AUG 2006 HIGHEST RN 904961-01-9 STRUCTURE FILE UPDATES: 28 AUG 2006 HIGHEST RN 904961-01-9 DICTIONARY FILE UPDATES:

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TSCA INFORMATION NOW CURRENT THROUGH June 30, 2006

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REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/ONLINE/UG/regprops.html

Uploading C:\Program Files\Stnexp\Queries\10535603\10535606tocopherol.str

chain nodes : 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 ring nodes : 1 2 3 4 5 6 7 8 9 10

chain bonds :

1-15 1-28 7-11 8-12 9-14 10-13 15-16 16-17 17-18 18-19 18-31 19-20

20-21 21-22 22-23 22-30 23-24 24-25 25-26 26-27 26-29

ring bonds :

1-6 1-2 2-3 3-4 4-5 4-10 5-6 5-7 7-8 8-9 9-10

exact/norm bonds :

1-6 1-2 2-3 3-4 5-6 9-14

exact bonds :

1-15 1-28 7-11 8-12 10-13 15-16 16-17 17-18 18-19 18-31 19-20 20-21

21-22 22-23 22-30 23-24 24-25 25-26 26-27 26-29

normalized bonds :

4-5 4-10 5-7 7-8 8-9 9-10

Match level :

1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:Atom 8:Atom 9:Atom 10:Atom

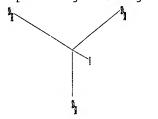
11:CLASS 12:CLASS 13:CLASS 14:CLASS 15:CLASS 16:CLASS 17:CLASS 18:CLASS

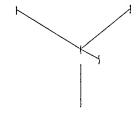
19:CLASS 20:CLASS 21:CLASS 22:CLASS 23:CLASS 24:CLASS 25:CLASS 26:CLASS

27:CLASS 28:CLASS 29:CLASS 30:CLASS 31:CLASS

L5 STRUCTURE UPLOADED

=> Uploading C:\Program Files\Stnexp\Queries\10535603\10535606methanetrisulphonate.str





chain nodes :
1 2 3 4 5
chain bonds :
1-4 2-4 3-4 4-5
exact bonds :
1-4 2-4 3-4 4-5

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS

L6 STRUCTURE UPLOADED

=> file caplus

COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 0.44 224.01

FULL ESTIMATED COST

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=> s 15 and 16

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

SAMPLE SEARCH INITIATED 15:15:20 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 147 TO ITERATE

100.0% PROCESSED 147 ITERATIONS 25 ANSWERS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**
BATCH **COMPLETE**

PROJECTED ITERATIONS: 2213 TO 3667 PROJECTED ANSWERS: 200 TO 800

L7 25 SEA SSS SAM L5

L8 65 L7

REG1stRY INITIATED
Substance data SEARCH and crossover from CAS REGISTRY in progress...
Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

SAMPLE SEARCH INITIATED 15:15:21 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 6 TO ITERATE

100.0% PROCESSED 6 ITERATIONS 0 ANSWERS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**
BATCH **COMPLETE**

PROJECTED ITERATIONS: 6 TO 266
PROJECTED ANSWERS: 0 TO 0

L9 0 SEA SSS SAM L6

L10 0 L9

L11 0 L8 AND L10

=> s 15

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

SAMPLE SEARCH INITIATED 15:16:54 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 147 TO ITERATE

100.0% PROCESSED 147 ITERATIONS 25 ANSWERS

SEARCH TIME: 00.00.03

FULL FILE PROJECTIONS: ONLINE **COMPLETE**
BATCH **COMPLETE**

PROJECTED ITERATIONS: 2213 TO 3667

PROJECTED TIERATIONS: 2213 TO 3007
PROJECTED ANSWERS: 200 TO 800

L12 25 SEA SSS SAM L5

L13 65 L12

=> s 16

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

SAMPLE SEARCH INITIATED 15:17:11 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 6 TO ITERATE

100.0% PROCESSED 6 ITERATIONS 0 ANSWERS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**

BATCH **COMPLETE**

PROJECTED ITERATIONS: 6 TO 266
PROJECTED ANSWERS: 0 TO 0

L14 0 SEA SSS SAM L6

L15 0 L14

=> file reg

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION

FULL ESTIMATED COST 1.84 230.37

FILE 'REGISTRY' ENTERED AT 15:19:17 ON 29 AUG 2006 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2006 American Chemical Society (ACS)

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STRUCTURE FILE UPDATES: 28 AUG 2006 HIGHEST RN 904961-01-9 DICTIONARY FILE UPDATES: 28 AUG 2006 HIGHEST RN 904961-01-9

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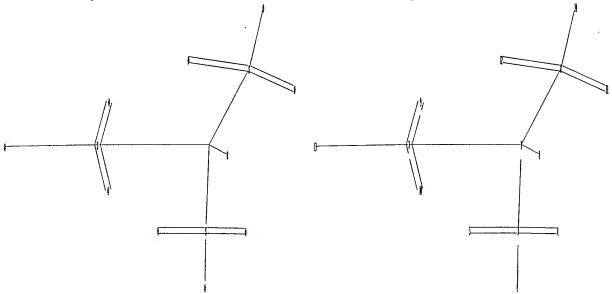
REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/ONLINE/UG/regprops.html

=>

Uploading C:\Program

Files\Stnexp\Queries\10535603\10535606methanetrisulphonatewithoutH.str



chain nodes :
1 2 3 4 5 6 7 8 9 10 11 12 13 14
chain bonds :
1-2 1-3 1-7 1-8 3-4 3-5 3-6 7-9 7-10 7-13 8-11 8-12 8-14
exact/norm bonds :
1-3 1-7 1-8 3-4 3-5 3-6 7-9 7-10 7-13 8-11 8-12 8-14
exact bonds :
1-2

Match level :

1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS 6:CLASS 7:CLASS 8:CLASS 9:CLASS 10:CLASS 11:CLASS 12:CLASS 13:CLASS 14:CLASS

L16 STRUCTURE UPLOADED

=> s 116

SAMPLE SEARCH INITIATED 15:19:31 FILE 'REGISTRY'

100.0% PROCESSED 1 ITERATIONS 0 ANSWERS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**

BATCH **COMPLETE**

PROJECTED ITERATIONS: 1 TO 80

PROJECTED ANSWERS: 0 TO 0

L17 0 SEA SSS SAM L16

=> s 116 ful

FULL SEARCH INITIATED 15:19:58 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 53 TO ITERATE

100.0% PROCESSED 53 ITERATIONS 15 ANSWERS

SEARCH TIME: 00.00.01

L18 15 SEA SSS FUL L16

=> d ibib hitstr 1-15

'IBIB' IS NOT A VALID FORMAT FOR FILE 'REGISTRY' 'HITSTR' IS NOT A VALID FORMAT FOR FILE 'REGISTRY'

The following are valid formats:

Substance information can be displayed by requesting individual fields or predefined formats. The predefined substance formats are: (RN = CAS Registry Number)

REG - RN

SAM - Index Name, MF, and structure - no RN FIDE - All substance data, except sequence data

IDE - FIDE, but only 50 names
SQIDE - IDE, plus sequence data

SQIDE3 - Same as SQIDE, but 3-letter amino acid codes are used

SQD - Protein sequence data, includes RN

SQD3 - Same as SQD, but 3-letter amino acid codes are used

SQN - Protein sequence name information, includes RN

CALC - Table of calculated properties EPROP - Table of experimental properties

PROP - EPROP and CALC

Any CA File format may be combined with any substance format to obtain CA references citing the substance. The substance formats must be cited first. The CA File predefined formats are:

ABS -- Abstract

APPS -- Application and Priority Information

BIB -- CA Accession Number, plus Bibliographic Data

CAN -- CA Accession Number

CBIB -- CA Accession Number, plus Bibliographic Data (compressed)

IND -- Index Data

IPC -- International Patent Classification

PATS -- PI, SO

STD -- BIB, IPC, and NCL

IABS -- ABS, indented, with text labels

IBIB -- BIB, indented, with text labels

ISTD -- STD format, indented

OBIB ----- AN, plus Bibliographic Data (original)

OIBIB ----- OBIB, indented with text labels

SBIB ----- BIB, no citations SIBIB ----- IBIB, no citations

The ALL format gives FIDE BIB ABS IND RE, plus sequence data when it is available.

The MAX format is the same as ALL.

The IALL format is the same as ALL with BIB ABS and IND indented, with text labels.

For additional information, please consult the following help messages:

HELP DFIELDS -- To see a complete list of individual display fields. HELP FORMATS -- To see detailed descriptions of the predefined formats. ENTER DISPLAY FORMAT (IDE):end

=> s 116 ful FULL SEARCH INITIATED 15:20:22 FILE 'REGISTRY' FULL SCREEN SEARCH COMPLETED - 53 TO ITERATE

100.0% PROCESSED 53 ITERATIONS 15 ANSWERS

SEARCH TIME: 00.00.01

L19 15 SEA SSS FUL L16

=> file caplus
COST IN U.S. DOLLARS

FULL ESTIMATED COST

SINCE FILE TOTAL
ENTRY SESSION
333.88 564.25

FILE 'CAPLUS' ENTERED AT 15:20:25 ON 29 AUG 2006
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=> s 119 L20 37 L19

=> d ibib hitstr abs 1-37

```
L20 ANSWER 1 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN
ACCESSION NUMBER: 2005:426557 CAPLUS
DOCUMENT NUMBER: 142:463900
TITLE: ACYTOTIC ACYTOTIC ACYTOTIC PROCESS FOR the pre-
                                                                                   Acylation process for the preparation of 2,3,5-trimethylhydroquinone diacylates in
  presence
                                                                                  of methanetrisulfonic acid catalyst
Aquino, Fabrice: Bonrath, Werner: Pr
DSM IP Assets B.V., Neth.
PCT Int. Appl., 10 pp.
CODEN: PIXXD2
  INVENTOR(S):
PATENT ASSIGNEE(S):
SOURCE:
  DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                               Patent
English
                                                                                NIND DATE APPLICATION NO.

A1 20050519 W0 2004-EP12058
AM, AT, AU, AZ, BA, BB, BG, BR, BM,
CU, CZ, DE, DK, DM, DZ, EC, EZ, EZ,
HR, HU, ID, IL, IN, IA, JP, KE, KG,
LT, LU, LV, MA, MD, MG, MK, NN, MM,
PG, PH, PL, PT, RO, BU, SC, SD, SE,
TR, TT, TZ, UA, UG, US, UZ, VC, VN,
KE, LS, MW, MZ, AA, SD, SL, SZ, TZ,
KZ, MD, RU, TJ/ TM, AT, BE, BG, CH,
FR, GB, GR, HJ, IE, IT, LU, MC, NL,
BF, BJ, CF, CG, CI, CM, GA, GN, GQ,
                  PATENT NO.
                                                                                                                                                                                                                             DATE
                   WO 2005044775
                                                                                                                                                                                                                             20041026
20041026
BZ, CA, CH,
FI, GB, GD,
KR, KZ, LC,
MZ, NA, NI,
SK, SL, SY,
ZA, ZM, ZW
ZM, ZW, AM,
CZ, DE, DK,
PT, RO, SE,
ML, MR, NE,
                                                                                                                                                                             EP12058
BR, BW,
EE, EG,
KE, KG,
MN, MW,
SD, SE,
VC, VN,
SZ, TZ,
BG, CH,
MC, NL,
GN, GQ,
                                                                                                                                                                                                        ES,
KP,
MX,
SG,
YU,
UG,
CY,
PL,
GW,
                                                                                                        20060719
                                                                                                           20060719 EP 2004-790843 20041026
ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
TR, BG, CZ, EE, HU, PL, SK
EP 2003-25513 A 20031107
                                                                                                                                                                                                                   W 20041026
                                                                                                                                                 WO 2004-EP12058
 OTHER SOURCE(S): CASREACT 142:463900
IT 54322-33-7, Methanetisulfonic acid
RL: CAT (Catalyst use): USES (Usea)
(acylation process for preparation of 2,3,5-trimethylhydroquinone
                 ylates
in presence of methanetrisulfonic acid catalyst)
54322-33-7 CABLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                   SO3H
  HO35-CH-SO3F
                 2,3,5-frimethyl-1,4-hydroquinone discylates (e.g., 2,3,5-trimethyl-1,4-benzquinone discetate), useful as intermediates in the preparation of a-tocopherol (no data), are obtained in high yield and selectivity by reacting 3,5-trimethyl-1,4-benzoquinone with an acylating agent (e.g., acetic anhydride) in the presence of catalytic quantities of
                     methanetrisulfonic acid.
                                                                                                     THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE
  REFERENCE COUNT:
  FORMAT
   L20 ANSWER 2 OF 37 CAPLUS COPYRIGHT 2006 ACS
ACCESSION NUMBER: 2004:965239 CAPLUS
DOCUMENT NUMBER: 141:395687
   DOCUMENT NUMBER
                                                                                  141:39568)
Process for the manufacture of tocyl and tocopheryl acystates
Sonrath, Werner: Haas, Alois: Hoppmann, Simone;
Netscher, Thomas: Pauling, Horat
DSM IP Assets B.V., Neth.
PCT Int. Appl., 15 pp.
CODEN: PIXXD2
Patent
   TITLE:
  INVENTOR (S):
   PATENT ASSIGNEE(S):
  DOCUMENT TYPE:
LANGUAGE:
                                                                                  English
1
   FAMILY ACC. NUM. CO
PATENT INFORMATION:
                                                       COUNT:
                  PATENT NO
                                                                                   KIND
                                                                                                         DATE
                                                                                                                                                 APPLICATION NO.
                                                                                                                                                                                                                            DATE
                            2004096 00 A1 20041111 W0 2004-EP4144 20040419
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
CR, CC, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
GE, GH, CM, HR, HU, ID, IL, IN, IS, PP, KE, KG, KP, KR, KZ, LC,
LK, IR, LS, IT, LU, LV, MA, MD, MG, MK, MM, MW, MX, MZ, NA, NI,
NO, NZ, OM, PE, PH, PL, PT, PO, RU, SC, SD, SE, SG, SK, SL, SY,
TJ, TM, TN, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZM,
BW, GH, GH, KE, DE, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE,
ES, FI, FR, GB, GR MU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI,
SK, TR, BF, BJ, CF CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
TD, TG

APPLN. INFO::
                  WO 2004096
              EP 2003-9522 A 2003
ER SOURCE(S): CASREACT 141:395687; MARPAT 141:395687
54322-33-7, Methanetrisulfonic acid
RL: CAT (Catalyst use) USES (Uses)
(process for manufacture of tocol and tocopherol acylates using perfluorosikylaulfonyl catalysts)
54322-33-7 CAPLUS
Methanetrisulfonic ccid (7CI, 9CI) (CA Yunsu
   PRIORITY APPLN. INFO.:
                                                                                                                                                                                                                 A 20030428
   OTHER SOURCE (S):
```

A process for the manufacture of tocyl acylate I [R = acyl; R1 = R2 = R3

503H | HO35-CH-SO3H

GI

L20 ANSWER 1 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN L20 ANSWER 2 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) or a tocopheryl acylate I (R = acyl; R5 = R7 = R8 = Me, R5 = H, R7 = R8 = Me, etc.) comprised reacting a corresponding tocol or tocopherol with an acylating agent in the presence of a catalyst of the general formula HCRIR2R3 (wherein R1, R2 and R3 each signify the sulfo group, or R1, R2 and R3 each signify a perfluoroally/sulfonyl group whereby at least two of R1, R2 and R3 are identical such perfluoroalkyl-sulfonyl groups, or R1 signifies the pentafluorophenyl sulfonyl group and R2 and R3 each signify an identical perfluoroalkylsulfonyl group). The main com. form of vitamin

E, being (all-rac)-\alpha-tocopheryl acetate I [R = acetyl; R5 = R7 = R8

= Me], can be manufd. by acytation of (all-rac)-\alpha-tocopherol
according to this process.

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS
RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

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L20 ANSWER 3 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 2004:823008 CAPLUS DOCUMENT NUMBER: 141:33463 TITLE: Crosslinked
                                                                              141:334863
Crosslinked polyoxyalkylene-polysiloxanes for use as nonaqueous salt-type electrolytes for lithium secondary batteries
Barrandon, Georges; George, Catherine; Vergelati, Caroll; Giraud, Yves
 INVENTOR (S):
 PATENT ASSIGNEE(S):
SOURCE:
                                                                              Rhodia Chimie, Fr.
Fr. Demande, 25 pp.
CODEN: FRXXBL
 DOCUMENT TYPE:
                                                                               Patent
  LANGUAGE:
 FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                 PATENT NO.
                                                                              KIND
                                                                                                    DATE
                                                                                                                                          APPLICATION NO.
                                                                                                                                                                                                                   DATE
                FR 2853321
FR 2853321
                                                                                                    20041008
20050506
20041021
                                                                                 A1
B1
                                                                                                                                          FR 2003-4153
                                                                                                                                                                                                                   20030403
                 WO 2004090037
                                                                                 A1
                                                                                                                                          WO 2004-FR708
                                                                                                                                                                                                                   20040323
                                                               A1 20041021 W0 2004-FR708 20040323
AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LS, LT, LU, LV, MA, MD, MG, MK, NN, MM, MX, MZ, NA, NT, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, CM, KE, LS, MM, MZ, SD, SL, SZ, TZ, UG, ZM, ZM, AM, AZ, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
                                       AE, AG,
CN, CO,
GE, GH,
                             w:
                                                    GH,
LR,
                                         LK,
NO,
TJ,
                                                    NZ.
               NO, NZ,
TJ, TM,
RW: BW, GH,
BY, KG,
ES, FI,
SK, TR,
TD, TG
                TD, TG

EP 160870S

A1 20051228 EP 2004-742318 20040323

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, L1, LU, NL, SE, MC, PT, 1E, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, FL, SK CN 1788054 A 20060614 CR 2004-8013072 20040323 A1TY APPLIN. INFO.:

FR 2003-4153 A 20030403
 PRIORITY APPLN. INFO.:
                                                                                                                                          WO 2004-FR708
                                                                                                                                                                                                        W 20040323
OTHER SOURCE(S):

MARPAT 141:334863

IT 54322-33-7DP, Methanetrisulfonic acid, derivs., salts
RL: DEV (Device component use): SPN (Synthetic preparation); TEM
(Technical or engineered material use); PREP (Preparation); USES (Uses)
(battery electrolytes containing; crosslinked
polyoxyalkylene-polysiloxanes
for use as nonaq. salt-type electrolytes for lithium secondary
batteries

RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                   SO3H
HO35-CH-SO3H
                Crosslinked polymeric electrolytes for lithium secondary batteries
   of: (1) a first poly(hydrogen organic siloxane) with ≥2 Si-H bonds per mol., (2) a second polysiloxane containing ≥2 Si-OH bonds per mol., (3)
L20 ANSWER 4 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN
ACCESSION NUMBER: 2004:823007 CAPLUS
TITLE: 141:334862 Photochemical crosslinked epoxy-polyoxyalkylene-polysiloxanes for use as nonaqueous salt-type electrolytes for lithium secondary batteries
INVENTOR(S): Gambut, Lucile; Vergelati, Caroll; Sanchez, Jean
                                                                              Alloin, Fannie
Rhodia Chimie, Fr.; Rhone Poulenc Chimie
Fr. Demande, 30 pp.
CODEN: FRXXBL
Patent
Prench
 PATENT ASSIGNEE(S):
SOURCE:
  DOCUMENT TYPE:
  LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                  PATENT NO.
                                                                                                    DATE
                                                                               KIND
                                                                                                                                          APPLICATION NO.
                                                                            ATE APPLICATION NO. DATE

A1 20041008 FR 2003-4150 20
B1 20050506
A2 20041021 WO 2004-FR707 20
A3 20050714
AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, LT, LU, LV, MA, MD, MG, MK, EM, MM, AX, FG, FH, FT, RO, RU, SC, SD, SE, SG, SK, TR, TT, TZ, UA, UG, US, UZ, VC, VM, YU, ZA, KE, IS, MM, MZ, SD, SI, SZ, TZ, UG, ZM, ZM, MD, RU, TJ, TM, AT, BE, BG, CH, CI, CZ, DE, GB, GR, MU, IE, IT, LU, MC, ML, PI, PT, RO, BJ, CF, CG, CI, CM, GA, CM, GQ, GW, ML, MR
                  FR 2853320
FR 2853320
WO 2004091033
WO 2004091033
                                                                                                                                                                                                                   20030403
                                                                                                                                                                                                                   20040323
WO 2004-PR707
                                                                                                                                                                                                         W 20040323
CTHER SOURCE(S): MARPAT 141:334862
IT 54322-33-7DP, Methanetrisulfonic acid, derivs., salts
RL: PRP (Properties); SPN (Synthetic prepyfation); TEM (Technical or engineered material use): PREP (Preparatyon); USES (Uses) (battery electrolytes containing; photochem. crosslinked epoxy-polyoxyalkylene-polysiloxanes/ for use as nonaq. salt-type electrolytes for lithium secondary batteries)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                   503H
 HO35-CH-503H
               Crosslinked polymeric electrolytes for lithium secondary batteries are prepared from a polyozyanosiloxane grafted with polyozyalkylene and epoxy functions, followed by radiochem. or photochem. (e.g., UV-Induced)
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L20 ANSWER 3 OF 37 CAPLUS COPYRIGHT—Z006 ACS on STN (Continued) a dehydrogenation-condensation catalyst, and (4) ≥1 salt electrolyte. The polyoxymixylene ether functions are derived from polyoxyethylene, polyoxymorpylene, or their mono-Me ethers. The dehydrogenation-condensation catalysts are typically metal complexes based on Pt, B, Rh, Ed, Sn, or Ir, preferably Karstedt (hydrosilylation) catalysts of formula IrCl(C:O) (PPh3)2. Suitable sait electrolytes ide catalysts of formula irci(C:O)(FFR3)2. Suitable sait electrolytes
Liclo4 LiBF4, LiAsF6, CF3SO3Li, LiN(CF3SO2)2, and LiN(C2FSSO2)2 in a
non-MG. electrolyte solvent, as well as other cations (e.g., transition
mefal cations, selected from Mn, Fe, Co, Ni, Cu, Zn, Ca, and Ag). Addnl.
dons include ammonium, amidinium, guanidnium cations, halides, ClO4-,
SCN-, BF4-, NO3-, AsF6-, PF6-, RSO3- (R = stearyl, CF3, octyl,
dodecylphenyl, and C16-perfluoroalkyl and -perfluoroaryl), (RSSO2)2N-,
and (R4SO2)(RSSO2)(R6SO2)C- (R4-6 = C1-6-perfluoroaryl) and -perfluoroaryl).
REFERENCE COUNT: THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

L20 ANSWER 4 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) crosslinking in the presence of a photosenatizer. The crosslinked epoxy-polyokyalkylene-polysiloxanes have the general formula R1xR2yR3ZSiO(4-x-z)/2, in which R1-3/er Cl-12-alkyl, C5-10-cycloalkyl, C6-10-aryl, aralkyl, or -OR4 (R4 -Mf. or Cl-15-alkyl; and x + y + z = 1-3).

The product polysiloxanes (typically prepd. from Si-H bond-contg. polysiloxanes by hydrosilylation grafting with allyl-terminated polyoxyalkylenes and epoxdes) are then combined with org. salts to form the battery electrolytes. Suitable salt electrolytes include Liclo4, LiBF4, LiAsF6, C73503ii, Lin(C73502)2, and LiN(C75502)2, other cations (e.g., transitiop metal cations); ammonium, amidinium, and guantidnium salts; org. subronates, imidodisulfonates, methanetrisulfonates, organoboron complex salts, ferricenium salts, and other onium salts (esp. iodonium and sulfonium compal.)

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

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L20 ANSWER 5 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN
ACCESSION NUMBER: 2004:453200 CAPLUS
DOCUMENT NUMBER: 141:23750
 DOCUMENT NUMBER:
 TITLE:
                                                                       Manufacture of a-tocopherol from
trimethyllydroquinone with isophy
presence of methane trisulfonate
Bornath Werner; Hoppmann, Simone
Netschfr; Thomas; Pauling, Horst
DSM IF Assets B.V., Neth
PCT fnt. Appl., 13 pp.
CODIN: PIXXD2
Pagent
English
                                                                          Manufacture of a-tocopherol from
                                                                                                                                                                                   reaction of
                                                                                                                                                                           ol or phytol in the
INVENTOR (5):
                                                                                                                                                                              Haas, Alois;
 PATENT ASSIGNEE(S):
SOURCE:
 DOCUMENT TYPE:
 LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                NO.
               PATENT NO.
                                                                         KIND
                                                                                           DATE
                                                                                                                               APPLICATION
                                                                                                                                                                                                  DATE
W1 2004046127
W1: AE, AG,
CO, CR,
GH, GH,
LR, LS,
OM, PG,
TTN, TR,
RW: GH, GH,
KG, K2
F1, FR
BP, B-
AU 2003270295
US 2006020139
PRIORITY APPLN. INFO
                                                                                 AFFLICATION NO.

20040603 W0 2003-EB10837 20030930
AT, AU, AZ, BA, BB, BC, GR, BY, BZ, CA, CH, CN, DE, DK, DM, DZ, BC, EE, EG, ES, FI, GB, GD, GE, LV, HA, MD, HG, HK, HM, HM, HK, MZ, NI, MO, NZ, PT, RO, RU, SC, SD, SE, SG, SK, SI, SY, TJ, TH, UA, UG, US, UZ, VC, VN YU, ZA, ZM, ZM WH, MZ, SD, SL, SZ, TZ, UG, ZM, ZM, AM, AZ, BY, TJ, TH, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, HU, IE, IT, LU, MC, NI, PT, RO, SE, SI, SK, TR, CI, CH, GA, GN, GQ, CM, ML, MR, NE, SN, TD, TG 20040615 AU 203-270295 2030930
20051123 CN 203-270295 2030930
EP 2005-2535603 203050519
EP 202-25990 A 20021121
                                                                       A1
AM,
CZ,
HU,
LU,
PL,
TZ,
LS,
RU,
GR,
CG,
A1
A
                                                                                                                                WO 2003-EP10837
                                                                                                                                                                                         W 20030930
OTHER SOURCE(S):

CASREACT 141:23750

IT 54322-33-7, Wethanetrisulfonic acid
RL: CAT (Catalyst use): USES (Uses)
(manufacture of a-tocopherol from the eaction of trimethylhydroquinone with isophytol or phytol in the presence of methane trisulfonate)

RN 54322-33-7 (CAPUIONATE)

CN Methanetrisulfonic acid (7CI, 9CI) (CA NDEX NAME)
                 SO3H
                 сн— so<sub>3</sub>н
               (all-rac)-o-tocopherol is prepared by the acid-catalyzed reaction of trimethylhydroquinone with isophytol of phytol in the presence of methane trisulfonate as the catalyst in an organic solvent.
 L20 ANSWER 6 OF 17 CAPLUS COPYRIGHT 2006 ACS on STN
                                                                                                                                                                           (Continued)
                                   - о- сн с г з
                                       î
                                                       CF3
               The invention concerns a method for preparing high yields of
 polyisocyanates
               containing urethione groups by cyclodimerization of isocyanate functions
 borne
               by initial monomer isocyanates. The invention is characterized in that
               comprises steps which consist in: (a) providing a reaction medium comprising initial monomer isocyanates, optionally in the presence of a solvent; (b) adding to said reaction mixture a (cyclo)dimerization
catalyst

comprising a compound of the family of super acids [such as (C3302)2NH];

(c) heating the reaction medium to a temperature ranging between 0°C and
300°C, advantageously between 20°C and 200°C until

the desired rate of transformation is obtained; (d) optionally,
inactivating or eliminating the reaction catalyst; and (e) optionally,
eliminating the unreacted monomer.

REFERENCE COUNT:

7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS
RECORD. ALL CITATIONS AVAILABLE IN THE RE
 catalyst
  FORMAT
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L20 ANSWER 6 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 2002:391791 CAPLUS DOCUMENT NUMBER: 136:386879
 DOCUMENT NUMBER:
TITLE:
                                                          136:386879
Method for catalytic dimerization of Bernard, Jean-Marie Rhodia Chimie, Fr. PCT Int. Appl., 42 pp. CODEN: PIXXD2
                                                                                                                                                  isocvanates
 INVENTOR (S):
 PATENT ASSIGNEE (S):
 SOURCE:
DOCUMENT TYPE:
                                                            Patent
     FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                          BI 20030829

AB 20020527 AU 2002-20807 20011120

A1 20031015 EP 2001-996571 20011120

CE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,

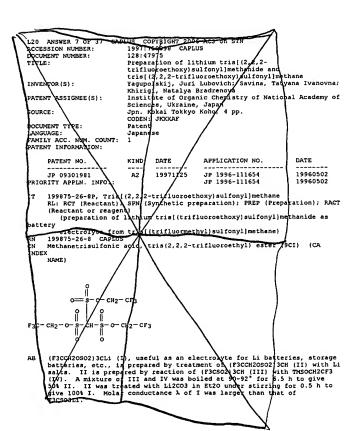
LV, FI, RO, MK, CY, AL, TR

A 20040106 BR 2001-15453 20011120

A1 20040122 US 2003-432050 20030519

FR 2000-14965 A 20001120
IE, SI
BR 2001015453
US 2004014970
PRIORITY APPLN. INF
                                                                                                        FR 2000-14966
                                                                                                                                                       A 20001120
                                                                                                        WO 2001-FR3648
                                                                                                                                                       W 20011120
OTHER SOURCE/S): MARPAT 136:386879

IT 426819-04-7, Tris[bis(trifluoromethyl)methoxysulfonyl]methane
RI: gAT (Catalyst use): USES (Uses)
dimerization of isocyanates in presence of super acids for manufacture of
uretdione-containing polyisocyanates)
RN 426819-04-7 CAPLUS
CN Methanetrisulfonic acid, tris[2,2,2-trifluoro-1-(trifluoromethyl)ethyl]
ester (9CI) (CA INDEX NAME)
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L20 ANSWER 8 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1997:69364 CAPLUS DOCUMENT NUMBER: 126:2211808 Investigations on the said accession.
                                                                                                                                                                                                                                                                                              L20 ANSWER 8 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                           (Continued)
                                                                                                                                                                                                                                                                                                                SO<sub>3</sub>H
                                                                                  Investigations on the acid-strength of
                                                                               Investigations on the acid-strength of
alkanepolysulfonic acids
Jueschke, Ralf; Sartori, Peter
Fachbereich 6 Anorganische chem., Gerhard-Mercator
Univ., Duisburg, D-47048, Germany
Zeitschrift fuer Naturforschung, B: Chemical Sciences
(1996), 51(12), 1691-1790
CODEN: 2NBSEN; ISSN: 0932-0776
Verlag der Zeitschrift fuer Naturforschung
Journal
                                                                                                                                                                                                                                                                                                                сн- so3н
                                                                                                                                                                                                                                                                                              HO35
  AUTHOR(S):
CORPORATE SOURCE:
  SOURCE:
                                                                                                                                                                                                                                                                                                     ●3/2 Ba
  PUBLISHER:
DOCUMENT TYPE:
                                                                                                                                                                                                                                                                                                              187610-57-7 CAPLUS
Methanetrisulfonic acid, trisilver(1+) salt (9CI) (CA INDEX NAME)
                                                                                Journal
   LANGUAGE:
                OMDE: German
St4322-33-7, Methanetrisulfonic acid
RL: PRP (Properties)
(preparation of silyl alkanepolysulfonates and estimation of acidity
                                                                                                                                                                                                                                                                                                                503H
  by
                                                                                                                                                                                                                                                                                                              -cн-sозн
                                                                                                                                                                                                                                                                                              HO35
                correlation with 29Si-NMR shift
                Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                   ●3 Ag(I)
                   503H
                                                                                                                                                                                                                                                                                                              187610-60-2P
                                                                                                                                                                                                                                                                                                               RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of sily alkanepolysulfonates and estimation of acidity
 HO35-CH-SO3H
                55110-91-3
RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation of sill alkanepolysulfonates and estimation of acidity
 IT
                                                                                                                                                                                                                                                                                                                        correlation with 2,551-NMR shift)
                                                                                                                                                                                                                                                                                                             Methanetrisulfonic
                                                                                                                                                                                                                                                                                                                                                                            cid, tris(trimethylsilyl) ester (9CI) (CA INDEX
                                                                                                                                                                                                                                                                                               NAME)
                correlation with 29Si-NMR shift) 55110-91-3 CAPLUS Methanetrisulfonic acid, tripotassi
                                                                             acid, tripotassium salt (3CI, 9CI) (CA INDEX NAME)
                   503H
 но<sub>3</sub>s-сн- sо<sub>3</sub>н
                                                                                                                                                                                                                                                                                               Me3Si-O
             ●3 K
                                                                                                                                                                                                                                                                                                                           acidity of alkanepolysulfonic acids was measured using the
                75533-77-6P 187610-57-7P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (preparation of silyl alkanepolysulfonates and estimation of acidity
                                                                                                                                                                                                                                                                                                              between the 29Si-NMR shift of the corresponding Me3Si ester of an acid
                                                                                                                                                                                                                                                                                                              the pKSl of the acid. The alkanepolysulfonates RCH(5035iMe3)2, RZ(5035iMe3)2, RC(5035iMe3)3, and Me35i035(CR12)s035iMe3 (R = H, F, Cl, Br, I; RI = H, F; n = 1-5) were prepared and studied.
                 correlation with 29Si-NMR shift) 75533-77-6 CAPLUS
                       thanetrisulfonic acid, barium salt (2:3) (9CI) (CA INDEX NAME)
L20 ANSWER 9 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1994:630311 CAPLUS
DOCUMENT NUMBER: 121:230311
Improvement of the synthesis of the tripotassium methanetrisulfonate monohydrate
AUTHOR(S): Sartori, Feter; Jueschke, Ralf
CORPORATE SOURCE: Fachbereich 6 Anorg. Chem., Und.v.-CH, Duisburg, Germany
Journal fuer Praktische Chemie/Chemiker-Zeitung
(1994), 336(4), 373-4
CODEN: JPCCBM: ISSN: 0941/1216
DOCUMENT TYPE: Journal
LANGUAGE: German
OTHER SOURCE(S): CASREACT 121:230311
T 54322-33-7, Methanetrisulfonic acid
RL: FMU (Formation, unclassified); RCT (Aeactant); FORM (Formation, nonpreparative); RACT (Reactant or reagent)
(improvement of the synthesis of the tripotassium methanetrisulfonate monohydrate)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                             L20 ANSWER 10 OF 37
ACCESSION NUMBER:
DOCUMENT NUMBER:
11:173247
The duodecet rule. Park 2
sulfonyl compounds
Robinson, E. A.
CORPORATE SOURCE:
SOURCE:
CORPORATE SOURCE:
DOCUMENT TYPE:
LANGUAGE:
L
                                                                                                                                                                                                                                                                                                                                                                                                                                                       2. C-H hydrogen bonding by
                                                                                                                                                                                                                                                                                                                                                                                                                                                   Toronto, Mississauga, ON, L5L
                                                                                                                                                                                                                                                                                                                                                                                                                                     55, 29-41
SN: 0166-1280
                                                                                                                                                                                                                                                                                                             UAGE: English
123177-61-7
RL: PRP (Properties)
  (total bond orders at sulfur in)
123177-61-7 CAPLUS
Methanetrisulfonic acid, ion(3-) (9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                503-
                                                                                                                                                                                                                                                                                               -03s-CH-so3-
                   ș03H
                                                                                                                                                                                                                                                                                                              By combining linear relationships between log k and log r, bond order (n) and force consts. (k) where k is the CS stretching force constant of a
  нозя-сн- возн
                                                                                                                                                                                                                                                                                                              of length r, a correlation is established between CS bond lengths and simple valence-bond bond orders. For a variety of S(VI) species with bonds to electroneg. ligands, including He groups, this is used to demonstrate that the total bond order at S in these species is six, consistent with a duodecet rule. Apparently, S-CH3 groups in compds.
               13224-87-0P
RL: SPN (Synthetic preparat
(improvement of the synthemonohydrate)
73224-87-0 CAPLUS
Methanetrisulfonic acid, tr
                                                                                                   tion); PREP (Preparation)
thesis of the tripotassium methanetrisulfonate
                                                                                                                                                                                                                                                                                                              as CH3SO2E/ CH3SO2OH, Me2SO2, and related species compete with other ligands for the delocalization of electron pairs into the valence shell
                                                                                       d, tripotassium salt, monohydrate (9CI) (CA INDEX
                                                                                                                                                                                                                                                                                                              the central sulfur atom, in this case through hyperconjugation. This accounts for the enhanced acidities of hydrogen atoms bonded to accarbon atoms in such compds., and their ability to form hydrogen bonds. Evidence for hydrogen bonding in CH3SO2F, CH3SO2Cl, and
                    SO3H
  нозя-сн- возн
                                                                                                                                                                                                                                                                                               (CH3) 250)
```

●3 K

● н2С

Thy Bagnall-Backer method for the preparation of the title compound [via decarboxylation of acetanilide in SO3/H2SO4 to give (HO3S)2C6H3NH2 and fc(SO3H)3 (I) then CaO and K2CO3 to give HC(SO3K)3·H2O (II)] is improved by the decarboxylation of acetone in SO3/H2SO4 to give I which with KOH gives 49% II.

and in CH3SO2OH (via both S-OH and S-CH3), is discussed. The use of C-H bynd lengths and stretching force consts. As a possible diagnostic tool

elect mols. capable of CH hydrogen bonding is also discussed.

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The electrochemistry of a dimeric and two monomeric cia-trioxomolybdenum(VI) complexes containing cyclic triamine ligands in protic and aprotic media: model compounds for the active site in formate

Herrmann, Willy; Wieghardt, Karl
Ruhr-Univ., Bochum, D-4630, Fed. Rep. Ger.
Polyhedron (1986), 5(1)21, 513-20

CODEN: PLYHDE: ISSN: 9277-5387

JOURNAL
                                                                                                  3-alkoxypropionates
Jones, Glenn C.; Nottingham, William/D.; Raynolds,
INVENTOR (S):
                                                                                                 Peter W.
Eastman Kodak Co., USA
 PATENT ASSIGNEE(S):
SOURCE:
                                                                                                                                                                                                                                                                                                                                                                dehydrogenase
                                                                                                 U.S., 4 pp.
CODEN: USXXAM
                                                                                                                                                                                                                                                                                                                                                                CORPORATE SOURCE:
DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                                                                                                                                                                                                                DOCUMENT TYPE:
                                                                                                                                                                                                                                                                                                                                                                                                                                                                 English
                                                                                                                                                                                                                                                                                                                                                                                    54322-33-7
                                                                                                                                                                          APPLICATION NO.
                                                                                                                                                                                                                                                                                                                                                                                   RL: PRP (Properties)
(electrochem. reduction of molybdenum trioxotriazacyclononane or
                    PATENT NO.
                                                                                                 KIND
                                                                                                                            DATE
                                                                                                                                                                                                                                                                      DATE
                                                                                                                                                                         US 1988-164663
CA 1989-591956
WO 1989-US763
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A1
A1
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19940118
                   US 4827021
CA 1326242
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trioxotrimethyltriazacyclononepe nuclear and dinuclear complexes on
                                                                                                                                                                                                                                                                       19890224
                    WO 8908636
                                                                                                                            19890921
                                                                                                                                                                                                                                                                       19890227
                                                                                                                                                                                                                                                                                                                                                                                    mercury in)
54322-33-7 CAPLUS
                   W: AU, JP, KR
RW: AT, BE, CH, DE, FR, GB, IT, LU
AU 8933504 A1 19891005
                                                                                                                                                                          NL, SE
AU 1989-33504
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                           OCI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                                                                                    Methanetrisulfonic acid (7CI,/
                                                                                                                                                                                                                                                                      19890227
                   AU 609288
EP 403528
EP 403528
                                                                                                                            19910426
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                                                                                                                             19901227
                                                                                                                                                                                                                                                                      19890227
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                                                                                                                            19930210
                   R: AT, BE, CH,
JP 03503282
JP 2738967
                                                                                               DE, FR,
T2
B2
                                                                                                                                                                             LU, NL, SE
JP 1989-503132
                                                                                                                            GB, 17,
19910725
                                                                                                                                                                                                                                                                                                                                                               HOSS-CH-SOSH
                                                                                                                                                                                                                                                                      19890227
                                                                                                                            19980408
19930215
19970104
                                                                                                                                                                                                                                                                                                                                                                                   The electrochem. of cis MoO3L (I), (L = N, N', N''-trimethyl-1,4,7-triazacyclononane) or cis-MoO3L' (II), (L' = 1,4,7-triazacyclononane in protic (CH3SO3H) and aprotic media (DMF) by cyclic voltammetry at hanging Hg drop (HMD) electrode. I and II are quasi-reversibly reduced to the control of the con
                                                                                                                                                                          AT 1989-903301
                   AT 85603
KR 9700138
                                                                                                                                                                                                                                                                       19890227
                                                                                                    В1
                                                                                                                                                                           KR 1989-72047
US 1988-164663
 PRIORITY APPLN. INFO.:
                                                                                                                                                                                                                                                                    19880307
                                                                                                                                                                                                                                                                                                                                                                                 hanging Hg drop (HMD) electrode. I and II are quasi-reversibly reduced a 2e-, 4H+ step in acidic solns. to give monomeric (LMo(IV)O(OH2)2]2+ and [L'MOO(OH2)2]2+, which in the case of I is further reduced to a monomeric Mo(III) species which then undergoes a reversible chemical reaction (dimerization). In contrast, reduced II undergoes chemical reactions (dimerization or trimerization of [L'MOO(OH2)2]2+ or comproportionation reactions)/dub to the fact that the unmethylated cyclic amine in II is sterically less demanding. I and II are redox-inactive in DMF (+1.0 to -1.7 V vs. Ag-AgCl at a Pt or a glassy C electrode). Dimeric [LZMO2O5] [ffF6]2, [III] is reversibly reduced to a Mo(V)2 dimer (2e-, 2H+ process) And, finally, to a Mo(III)2 dimer (4e-, 4H+) in 0.1M CH3SO3H at an HMD electrode, contrasting with the behavior in aprotic medium (DMF), Pt electrode), where 2 reversible 1-electron transfer processes
                                                                                                                                                                          EP 1989-903301
                                                                                                                                                                                                                                                         A 19890227
                                                                                                                                                                          WO 1989-US763
                                                                                                                                                                                                                                                         A 19890227
                 R SOURCE(S):
MARPAT 11:153222
54322-33-7, Methanetrisbulfonic acid
RL: CAT (Catalyst uses): USES (Uses)
(catalyst, for addition of alkoxymethane with ketene)
54322-33-7 CAPLUS
Methanetrisulfonic/acid /7CI, 9CI) (CA INDEX NAME)
OTHER SOURCE(S):
                      șo3H
                                                                                                                                                                                                                                                                                                                                                                generating/

the/blue, mixed-valence species Mo(VI)/(V) and an Mo(V)2 dimer were

observed
HORS-CH-SORH
                   R1OCH2CR3R4CO2R2 (I; R1, R2 = C1-8 alkyl; R3, R4 = H, alkyl, aryl),
useful as solvents in the formation of coating compns., are prepared by addition of R10CH2OR2 with R3R4C:CO in the presence of CH2(SO3H)2, CH(SO3H)3, or
mixture
thereof. A mixture of 23.0 g CH2(OMe)2, 0.071 CH2(SO3H)2, and 0.30 and
CH2:CO was stirred at 25-40 under N to give 24.9 g I (R1 = R2 =
Me, R3 = R4 = H). Similarly prepared was I (R1 = R2 = Et, R3 = R4 = H).
L20 ANSWER 13 OF 37
ACCESSION NUMBER:
DOCUMENT NUMBER:
1985:184717 CAPLUS
102:184717 CAPLUS
102:184717 CAPLUS
102:184717 CAPLUS
102:184717 CAPLUS
104:184717 CAPLUS
107:184717 CAPLUS
107:184717
                                                                                                                                                                                                                                                                                                                                                                L20 ANSWER 14 OF 37
ACCESSION NUMBER:
DOCUMENT NUMBER:
1983:523106 CAPLUS
99:123106
N-phenylcarbamate ester oligomers
Asshi Chemical Industry Co., Ltd.
Jpn. Kokai Tokkyo Koho, 6 pp.
CODEN: JKXXAF
Patent
Patent
                                                                                                                                                                                                                                                                                                                                                                                                                                                                Patent
Japanese
                                                                                                                                                                                                                                                                                                                                                                 DOCUMENT TYPE:
                                                                                                                                                                                                                                                                                                                                                                 DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                 Patent
English
 LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
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                                                                                                                                                                                                                                                                                                                                                                                                                                                                                            DATE
                    PATENT NO.
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                                                                                                                                                                           APPLICATION NO.
                                                                                                                                                                                                                                                                       DATE
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JP 62008430
PRIORITY APPLN. INFO.:
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19870223
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AU 538363
AU 8171460
PRIORITY APPLN. INFO.:
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                                                                                                                            19840809
19811217
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                                                                                                                                                                           AU 1981-71460
                                                                                                                                                                                                                                                                       19800613
                                                                                                                                                                                                                                                                                                                                                                                  54322-33-7
RL: CAT (Catalyst use); USES (Uses) (catalysts, for oligomerization methylene donors) 54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI)
                   D4322-33-7
RL: CAT (Catalyst use); USES (Uses)
(catalyst, for alkoxylation of alc
54322-33-7 CAPUE)
Hethanetrisulfonic acid (7CI, 9CI)
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                       phenylcarbamate esters with
                                                                                                                                                                         . and phenols)
                                                                                                                                                              (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                (CA INDEX NAME)
                      șo3H
                                                                                                                                                                                                                                                                                                                                                                                      503H
                    - ch- so3H
HO3S
                                                                                                                                                                                                                                                                                                                                                                                  N-Phenylcarbamate eaters are oligomerized with methylene donors in the presence of solid-sypported polysulfonic acids. Thus, 10g kieselguhr and 3.5g CH2 [SO3H)2 [7] [503-40-2] in 10 mL H2O was dried and baked 6 h at 200° to give catrlyst containing 14% I. A mixture of the above catalyst 3, Et N-phenylcarbamate (II) 8.25, and dimethoxymethane 1g in 40 mL sulfolane was heated 1 h in an autoclave at 140° to convert 56% II to oligomers/ (87093-19-4) containing 73% 1,1°-methylenebis(4-ethoxycarbonylaminobenzene) [10097-16-2] and 11% 1,3°-methylenebis(4-ethoxycarbonylaminobenzene) [70381-86-1].
                 Alcs. and phenols were alkoxylated with alkylene oxides in the presence
                    catalysts consisting of organic sulfonic acids or their salts. The
                   onic
acids chosen have an acid strength greater then that of methanesulfonic
acid; the cations are chosen from first row transition metals, Be, Mg,
                  B, Al, Sn, or 2r. Thus, BuOH was ethoxylated with ethylene oxide (6:1
                   ratio) at 160° in the presence of 2n methanesulfonate to give 93.8% ethylene glycol monobutyl ether and small amts. of diethylene glycol monobutyl ether/ (20.9:1 mol ratio). Other catalyst used were Zn benzenesulfonate and Al methanedisulfonate.
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L20 ANSWER 12 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1986:431749 CAPLUS DOCUMENT NUMBER: 105:31749

nonane) was

DATE

19811118

19811118

TITLE:

L20 ANSWER 11 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1989:553222 CAPLUS DOCUMENT NUMBER: 111:153222

TITLE:

Process for the preparation of alkyl

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L20 ANSWER 15 OF 37
ACCESSION NUMBER:
DOCUMENT NUMBER:
1983:504727 CAPLUS
99:104727
Methanetriaulfonic acid derivatives
AUTHOR(5):
CORPORATE SOURCE:
SOURCE:
CORPORATE SOURCE:
                                                                                                                                                                                                                                                                                                                                                                           L20 ANSWER 16 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1981:174556 CAPLUS DOCUMENT NUMBER: 94:174556 CAPLUS 2-8ydcoxyacetophenone Aia Fried
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                 AUTHOR(S):
CORPORATE SOURCE:
 DOCUMENT TYPE:
                                                                                                     Journal
                                                                                                                                                                                                                                                                                                                                                                           SOURCE:
                                                                                                    Russian
CASREACT 99:104727
 LANGUAGE: RUSSIAN
OTHER SOURCE(S): CASREACT 99:104727
IT 86107-38-2P
RL: SPN (Synthetic preparation); PREP (Preparation)
                                                                                                                                                                                                                                                                                                                                                                            DOCUMENT TYPE:
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                  English
CASREACT 94:174556
                                                                                                                                                                                                                                                                                                                                                                           LANGUAGE: English
OTHER SOURCE(S): CASREACT 94:174556
IT 54322-33-7
RL: CAT (Catalyst use); USES (Uses)
(catalysts, for Fries rearrangement of Ph acetate)
RN 54322-33-7 CAPLUS
                     (preparation of)
86107-38-2 CAPLUS
                     Methanetrisulfonic acid, triphenyl ester (9CI) (CA/INDEX NAME)
                                                                                                                                                                                                                                                                                                                                                                                                 54322-33-7 CAPLUS /
Methanetrisulfonic acid (//CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                                                                                                   503H
                   0
                                          ů
                                                                                                                                                                                                                                                                                                                                                                            HOAS-CH-SOAH
                              CH-
                                                                                                                                                                                                                                                                                                                                                                                              PhoAc was prepared directly from PhOH and AcOH in the presence of H2SO4; thermal rearrangement of PhOAc was studied in the presence of methanesulfonic and -trisulfonic acids, p-mc6H4SO3H, and sulfonated cation exchange resin. Catalysis by H2SO4, H2SO4, H2SO4, H2O, H3PO4, NaHSO4, and Cacl2 was also studied. The catalysts gave o- and p-HOC6H4Ac, with
                     54322-33-7
                    RL: RCT (Reactant); RACT (Reactant or reagent)
(reaction of, with sulfur tetrafluoride)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                                                                                                amts. of the ortho derivative at higher temps. The rates, product
                                                                                                                                                                                                                                                                                                                                                                            ratios, and yields of these processes were compared. AlCl3, ZnCl2, and BF3 all gave faster reactions than the proton acids, and higher ortho-para ratios, especially
                         SO3H
                                                                                                                                                                                                                                                                                                                                                                                                mith By3 at low temps. The results obtained with the proton acids and other/compds. and with the Lewis acids were compared. The object of the studyes was to provide a com. route to pyrocatechol.
 HO35-CH-SO3H
                    CH(SO3H)3 and SF4 gave CH(SO2F)3, which with Br, iodine, or Me3SiCl gave the corresponding RC(SO2F)3. CH2(SO2F)2 and Ag2O gave AgCH(SO2F)2, which with Me1 or CyfSSCl gave, resp., MeCH(SO2F)2 and Cef5SCH(SO2F)2.
4-MeC6H4SO3F/and trifluoromorpholinosulfur gave 4-MeC6H4SO2F.
L20 ANSWER 17 OF 37
ACCESSION NUMBER:
DOCUMENT NUMBER:
1981:139146 CAPLUS
94:139146
Tris(fluorosulfonyl)methane, HC(SO2F)3
Kloeter, Gerhard: Britzkow, Hanar Seppelt, Konrad
Anorg. Chem. Inst. Univ. Heidelberg, D-6900/1, Fed. Rep. Ger.
Angewandte Chemie/ (1980), 92(11), 954-5
CODEN: ANCEAD: ISSN: 0044-8249
Journal
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ANSWER 17 OF 37 CAPLUS COPYRIGHT 7006 ACS on STN (Continued) of the K and Rb salts were detd. Freatment of AgC(SO2F)3 with MeI or X2 gave, resp., MeC(SO2F)3 and XC(SO2F)3 (X = Cl, Br, I). FC(SO2F)3 was prepd. by fluorinating I with Xef2.

via Fries rearrangement and comparative applied study

● K Successive treatment of HC(503K)3 with Ba2+, H2S04, and SF4 gave, via HC(503Bx/2)3 and HC(503H)3, HC(502F)3 (I), which is a monobasic acid with strength felling between those of HS03F and HN03. Treatment of I with bases gave the salts MC(502F)3 (M = K, Rb, Cs and Ag); crystal structures

/3333-/6-3
RE: RCT (Reactant): RACT (Reactant or reagent)
(reaction of, with barium ion)
75333-76-5 CAPLUS
Methanetrisulfonc acid, monopotassium salt (9CI) (CA INDEX NAME)

DOCUMENT TYPE: LANGUAGE: IT 54322-33-7P

SO3H HO35-CH-SO3H 75533-77-6P

SO3H HO35-CH-503H

> ●3/2 Ba 75533-76-5

SO3H нозя-сн- возн

Journal German

(Reactant or reagent)

(Preparation and reaction of, with sulfuric acid)
75533-77-6 CAPLUS

Methanetrisulfonic acid, barium salt (2:3) (9CI) (Ci

54322-33-7P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)
(preparation and reaction of, with sulfur tetrafluoride)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT

barium salt (2:3) (9CI) (CA INDEX NAME)

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L20 ANSWER 18 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1981:68557 CAPLUS DOCUMENT NUMBER: 94:68557
                                                                                         94:68557

New electrolytes for direct methane fuel cells
Brummer, S. B.; Foos, J.; McHardy, J.; McVeigh, J.;
Toland, D.; Turner, M.
EIC Corp., Newton, NA, USA
Report (1979), DOE/ET/11321-11/74 pp. Avail.: NTI:
From: Energy Res. Abstr. 1980/5(10), Abstr. No.
TITLE:
AUTHOR (S):
CORPORATE SOURCE:
                                                                                                                                                                                                          74 pp. Avail.: NTIS
5(10), Abstr. No.
15977
DOCUMENT TYPE:
                                                                                          Report
English
                  54322-33-7
                   RL: USES (Uses)
                             (electrolytes, for direct-methane fuel celes, development and testing of)
                 of)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                    ŞO3H
HOSS-CH-SOSH
                 A program aimed at developing a fuel-dell electrolyte for the direct oxidation of CH4 and/or impure H fuels is described. Of interest are
               oxidation of CN4 and/or impure H fuels is described. Of interest are and tribasic methanesulfonic acids CX2($03H)2 and CX($03H)3 where X is H, F, or Cl. Synthetic routes to CN2($03H)2 [503-40-2], CN($03H)3 [532-43-7], CC12($03H)2 [76371-35-2], and CC1($03H)3 [76371-36-3] were identified and optimized. The diphenyl ester of C72($03H)2 was prepared and various approaches to CF($03H)3 [76371-37-4] were investigated. In parallel with the synthetic program, apparatus was designed and fabricated for the festing of the electrolytes under fuel-cell conditions. A new PTFE test cell was developed for testing small amts. of electrolyte. Electrodes with low Pt loading were loped for use in electrolyte evaluation. Optimum performance with H3P04 was achieved using 1 mg Pt/cm2 and 1 mg TFE 30/cm2 deposited on TFE tape, supported on a Au-plated Ta screen, and sintered. Preliminary half-cell tests using CX2($03H)2 and Cf($03H)3 indicated that these acids are insufficiently stable for use as fuel-cell electrolytes. However, tests using CC12($03H2) and CC1($03H3) indicated that these acids are insufficiently stable for use as fuel-cell electrolytes. However, tests using CC12($03H2) and CC1($03H3) avere encouraging, yielding H oxidation
rates equal to or better than those using H3PO4. Stability tests were conducted
conducted
by heating a sample of each acid at 130° for 30 days under N, O, and H. At the end of the test, each sample was analyzed for decomposition in no case did IR anal. indicate significant decomposition and in the case
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L20 ANSWER 19 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN (Continued)
.apprx.113' indicate an expansion from tetrahedral stereochem. and
the S-C bond lengths of [apprx.1.81 Å are longer than those for
K2(CH2(SG3)2) (1.77 Å) and Ca(CH3SO3)2 (1.75 Å). The IR (4000 to
50 cm-1) and Raman spectra of K3(CH(SO3)3]. H20 and K3(CD(SO3)3]. D20 were
detd. at 77' K and interpreted in accordance with the structure,
and with the use of the spectra of anhyd K3(CH(SO3)3] and of aq. solns.
of the sol. Li salts. The SO3 groups show their characteristic group
frequencies: veymic-S) 762, vasym(C-S) 820, 8aym(CS3) 170, and
Basym(CS3) .apprx.210 cm²1, resp. IR spectra of samples contg. the
isotopically dil. HDO species confirm the presence of 2 types of H bond
per H20 mol.

of the Chloroacids, only a trace amount of free Cl- was observed Conductivity

chloroacids, only a trace amount of the same conductivity as aqueous showed the aqueous scids to be of the same conductivity as aqueous H3PO4. The dihydrate of CM2(SO3H)2 was more conductive than CF3SO3H.H2O [49789-04-0] but less conductive than 100% H3PO4.

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L20 ANSWER 19 OF 37 CAPLUS COPYRIGHT 2006 ACS on SYN
ACCESSION NUMBER: 1980:145976 CAPLUS
DOCUMENT NUMBER: 92:145976
TITLE: 92:145976
Crystal structure, and the infrared and Raman
                                                         of tripotassium methanetrisulfonate hydrate,
K3[CR(803)3].H20
Hall, John R.; Johnson, Robert A.; Kennard, Colin H.
L.; Smith, Graham
Dep. Chem., Univ. Queensland, Brisbane, 4067,
Australia
Journal of the Chemical Society, Dalton Transactions:
Inorganic Chemistry (1972-1999) (1980), (1), 149-55
CODEN: JCDTBI; ISSN: 0300-9246
 spectra.
AUTHOR (S):
CORPORATE SOURCE:
SOURCE:
DOCUMENT TYPE:
                                                         Journal
English
 LANGUAGE:
            73224-78-9
            RL: PRP (Properties)
           (IR and Raman spectra of aqueous)
73224-78-9 CAPLUS
Methanetrisulfonic acid, trxllithium salt (9CI) (CA INDEX NAME)
             503H
HO35
            -сн- soaн
        ●3 Li
            73224-87-0
RL: PRP (Properties)
            KE: YEV (Properties)
(crystal and mol. structure, and IR and Raman spectra of)
73224-87-0 CAPUS
Methanetrisujtonic acid, tripotassium salt, monohydrate (9CI) (CA INDEX
             ŞO 3H
            - CH- SO:
         ●3 к
         ● H2O
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The crystal and mol. structure of K3(CH(SO3)3).H2O was determined by diffraction as an aid in the assignment of the IR and Raman spectra. The structure was solved using direct methods and refined by full-matrix squares to R 0.045 for 839 observed unique reflections. The S-C-S angles of

L20 ANSWER 20 OF 37
ACCESSION NUMBER:
DOCUMENT NUMBER:
2:139193 CAPLUS
1975:139193 CAPLUS
2:139193 Reactions of dipotassium diazomethanedisulfonate in aqueous solution
AUTHOR(S):
CORPORATE SOURCE:
DOCUMENT TYPE:
LANGUAGE:
COMPART SOURCE (S):
DOCUMENT TYPE:
LANGUAGE:
COMPART SOURCE (S):
COMPART SOURCE (S):
COMPART SOURCE (S):
CAPPUS COPYRIGHT 2006 ACS on STN
1975:139193 CAPPUS
Reactions of dipotassium diazomethanedisulfonate in aqueous solution
Young, J/Michael
Dep. Phirmacol., Univ. Cambridge, Cambridge, UK
Journal of the Chemical Society, Perkin Transactions
1: Organic and Bio-Organic Chemistry (1972-1999)
11976, 1221, 2241-3
CODER:
JOURNAL JOUR CODEN: JCPRB4; ISSN: 0300-922X
JOYANA
DOCUMENT TYPE: JOYANA
LANGUAGE: English
OTHER SOURCE(S): CASREACT 82:139193
T 55110-91-3P
RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)
RN 55110-91-3 CAPLUS
CN Methanetrisulfonic acid, tripotassium salt (3CI, 9CI) (CA INDEX NAME)

503H HO35-CH-503H

●3 к

Decomposition of (KO3S)2CN2 (I) in H2O at 4° gave (KO3S)3CH and (KO3S)2C:NNH2, in N KOH at 70° gave (KO3S)2C:NNKSO3K, and in aqueous piperidine at 70° gave (KO3S)2CH2. I with HeOH gave (KO3S)2CH0Me.

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L20 ANSWER 21 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1975:64946 CAPLUS
DOCUMENT NUMBER:
TITLE:
INVENTOR(S):
                                                        Solid catalysts for heterogeneous reactions
                                                      RODA, PECER
IMI (TAMI) Institute for Research and Development
Ger. Offen., 21 pp.
CODEN: GWXXBX
 PATENT ASSIGNEE (S):
DOCUMENT TYPE:
 FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                 APPLICATION NO.
           PATENT NO.
                                                      KIND
                                                                      DATE
                                                                                                                                                    DATE
                                                                                                DE 1974-2401958
US 1974-430804
GB 1974-1839
JP 1974-7615
                                                                      19740718
19751118
19760818
           DE 2401958
US 3920582
GB 1446964
JP 50046587
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A
A
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                                                                                                                                                     19740115
PRIORITY APPLN. INFO.:
                                                                                                IL 1973-41330
         54322-33-7
   T 9432-33"
RL: RCT (Reactant): RACT (Reactant or reagent)
(catalyst preparation with for heterogeneous catalysis or organic
eactions)
         54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
             șo3H
нозя-сн- 503н
           Catalysts for heterogeneously catalyzed reactions were prepared by impregnation of cafriers with sulfonic acids. Thus, 50 g SiO2-Al203 pellets were treated for 30 min with 14 g benzene-1,3-disulfonic acid in H20 at 80°, dried for 6 hr at 150°, and calcined for 6 hr at 200° to give 60/g catalyst. A H20-C2H4 mixture of mol. ratio 1:1 was passed over this catalyst at 195° to give a C2H4-C2H509 conversion of 0.3-0.5 mole % without splitting off acid from this catalyst.
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L20 ANSWER 23 OF 37 CAPLUS COPYRIGHT 2006 ACS
ACCESSION NUMBER: 1963:14557 CAPLUS
DOCUMENT NUMBER: 58:14557
ORIGINAL REFERENCE NO.: 58:2371g-h
                                                                                             Touey, George P.; Goins, Rex H.
Eastman Kodak Co.
   INVENTOR (S):
 INVENTOR (S):
PATENT ASSIGNEE (S):
SOURCE:
DOCUMENT TYPE:
LANGUAGE:
PATENT INFORMATION:
                                                                                             3 pp.
Patent
Unavailable
                                                                                                                                                                    APPLICATION NO.
                    PATENT NO.
                                                                                             KIND
                                                                                                                                                                                                                                                           DATE
                  US 3053884 19620911 US 1959-8453
54322-33-7, Methanetrisulfonic acid
(as catalyst for esterification)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                       SO<sub>3</sub>H
  нозя-сн- возн
                   CH2(SO3H)2 and CH(SO3H)3 are superior catalysts for preparing esters by treating saturated aliphatic mono- and polyhydroxy alcs. with phenyl dicarboxylic acids or saturated aliphatic carboxylic acids and their annydrides. A/lower concentration of catalyst is required and the ester
produced
is nearly colorless and is heat stable. Two moles phthalic anhydride and
five moles BuOH were refluxed 7 hrs. in the presence of various acid
catalysts. The catalyst used, the catalyst concentration based on the
phthalic
anhydride, and the percent phthalic acid in the product are: CH2(SO3H)2,
0.1, 0.02; CH(SO3H)3, 0.1, 0.03; H2SO4, 0.1, 0.35; MeSO3H, 0.2, 1.6;
MeCGH4SO3H, 1.0, 2.0; (CH2SO3H)2, 0.2, 0.85. Data are given which show
the superiority of these two catalyst for the esterification of n-octyl
alc. with adipic acid and glycerol with 2-ethylhexanoic acid.
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L20 ANSWER 24 OF 37 CAPLUS COPYRIGHT 2006 AGS on STN
ACCESSION NUMBER: 1953:70596 CAPLUS
DOCUMENT NUMBER: 47:70596
ORIGINAL REFERENCE NO.: 47:11919f-i
TITLE: Acid-base equalibrya in glacial acetic acid
AUTHOR(S): Smith, Thor L.: Eliott, John H.
CORPORATE SOURCE: Hercules Powder Co., Wilmington, DE
Journal of the American Chemical Society (1953), 75,
3566-71
CODEN: JACSAT/ ISSN: 0002-7863
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable
IT 54322-33-7, Methanetrisulfonic acid
(ionization in AcoN)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                              SO3H
  нозя-сн- возн
  AB Values of Ho for dilute solns. (5 + 10-4 to 5 + 10-3 M) of 11 strong acids in AcoM containing 0.12% water were measured by use of indicators

--naphtholbenzein/(I) and o-nitroaniline. Ho = -log(BH+)/(B) + pKa, where (BH+) and (B) are the concns. of the acidic and basic forms of an indicator, and pKa is the thermodynamic dissociation constant for the conjugate
   conjugate acid of the indicator. The pKa for I was evaluated as 0.53. The order
                          increasing scid strength at equal molarities is: HCl, methanesulfonic, sulfuric, carboxymethanesulfonic, chloromethanesulfonic, chloromethanesulfonic, chloromethanesulfonic, molaromethanedisulfonic, chloromethanedisulfonic, and methanetrisulfonic acids. Ho values for anhydrous solns of 4 monobasic acids at 5 * 10-3 M were measured, and from the increased acidity found, equilibrium consts. for the reaction
from the increased acidity found, equilibrium conditions of the acids with water were calculated H2SO4 was found to be monobasic. Dissociation consta. Kc, of HCl, HBr, HClO4, and H2SO4 in AcOH (calculated from conductivity data of Kolthoff and Willman (C.A. 28, 3644.1)) are 5.1 + 10-10, 1.9 + 10-7, 9 + 10-7, and 7.4 + 10-9, resp. The fact that values of ApKe from conductivity and from Ho data are in reasonable agreement shows that equilibrium in AcOH involve, primarily, undissocd. species.
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L20 ANSWER 22 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1963:468923 CAPLUS DOCUMENT NUMBER: 59:68923 CAPLUS ORIGINAL REFERENCE NO: 59:12707d-f

KIND

US 3082258 19630319 54322-33-7, Methanetrisulfonic acid (catalyst in alkylation of phenols) 54322-33-7 CAPLUS

weight methanetrisulfonic acid trihydrate.

Alkylation of phenoi McConnell, Wayne y.;

DATE

AB The preparation of 2,6-di-tert-butyi-q-methylpheno!

4-methylpheno!

(II) and isobutylene using hydrated methanedi- or trisulfonic acid catalyfis was described. I was useful as an antioxidant and stabilizer for fats and oils. Thus, 112 g. isobutylene was bubbled through a flass containing 108 g. II and 1.1 g. methanedisulfonic acid dihydrate (III)

10 100

of benzene. In the initial stages the temperature varied from 25-40° cf. benzene. In the initial stages the temperature varied from 25-40° due to the cooling effect of isobutylene refluxing in a dry ice-actor cooled condenser. Thereafter the temperature was held at 40° for a total reaction time of 6 hrs. The supernatant liquid was decanted from the catalyst. Unreacted II (61) and 2-tert-butyl-4-emethylphenol (31% conversion) were extracted with aqueous NaOH. After removal of C6H6, I

obtained (63% conversion), m. 68-9° (50% aqueous NeOH). Under the same conditions, 5.5 g. III gave an 88% conversion to I. Only a 20% resunt resulted from use of 1,2-ethanedisulfonic acid. Benzenedisulfonic acid caused polymerization of isobutylene. When p-toluenesulfonic acid or H2804 was used in concentration of 5% based on the weight of 11 the poorer color and odor. White, odorless I could also be prepared in 84 80% conversions, resp., using 2.2 g. III and no solvent or using 1% by

The preparation of 2,6-di-tert-butyl-4-methylphenol (I) from hylphenol

Eastman Kodak Co

Unavailable

Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

V.; Davis, Herman E.

APPLICATION NO.

us 1960-28557

DATE

19600512

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: TITLE:

PATENT ASSIGNEE (S):

PATENT INFORMATION: PATENT NO.

INVENTOR(S):

IТ

DOCUMENT TYPE: LANGUAGE:

șo3H

HO35-CH-SO3H

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L20 ANSWER 25 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN

ACCESSION NUMBER: 1952:26486 CAPLUS

DOCUMENT NUMBER: 46:26486
ORIGINAL REFERENCE NO.: 46:4471e-g

TITLE: The synthesis of certain derivatives of methionic acid
AUTHOR(S): Shu, Rue-Cheng; Jenkins, Glenn L.; Christian, John E.
CORPORATE SOURCE: Purdue Univ., Lafayette, IN
SOURCE: Journal of the American Pharmaceutical Association
(1912-1977) (1951), 40, 86-8/
CODEN: JPHRAJ: ISSN: 0003-0465

DOCUMENT TYPE: Journal
LANGUAGE: Unavailable
IT 873390-12-6, Nethanetrisulfonic acid, triethyl ester
(preparation of)
RN 873390-12-6 CAPLUS
CN Methanetrisulfonic acid, triethyl ester (5CI) (CA INDEX NAME)

ETC S-CH-S-OET

O O

AB N,N'-Di-p-phenetylmethionamide, m. 219.2-21*, was prepared by adding
CM2(SO2CI)2 in anhydrous C6H6 dropwise to p-phenetidine in C6H6 at 0*,
refluxing, cooling, filtering out the precipitate, washing with cold H2O,
decolorizing with charcoal, and recrysto, from aic. dioxane. By similar
procedure were prepared the following analogs: N,N'-di-o-phenetyl, m.
131-2*; N,N'-di-m-phenetyl, m. 166.8-7.8*,
N,N'-bis(p-dziebobutoxyphenyl)methionamide, m. 180.5-1.0*.
Methionate/ esters di-l-naphthyl, m. 136.8-7.4*; di-2-naphthyl, m.
231.5-6.17; bis(o-methoxyphenyl), m. 99.6-100.4*.
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L20 ANSWER 27 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1951:59987 CAPLUS
DOCUMENT NUMBER: 45:59987
ACTION NUMBER: 45:59987
ACTION REFERENCE NO.: 45:10190b-d
Ethyl methanetrisulfonate and its hydrolysis
Samen, Evald
CORPORATE SOURCE: Univ. Uppsale, Swed.
SOURCE: COEN: ARKEAD; ISSN: 0365-6128
DOCUMENT TYPE: Unavailable
IT 873390-12-6, Methanetrisulfonic acid, triethyl ester
(and its hydrolysis)
RN 873390-12-6, CAPLUS
CN Methanetrisulfonic acid, triethyl ester
(and its hydrolysis)
CN Methanetrisulfonic acid, triethyl ester (5CI) (CA INDEX NAME)

AB cf. C.A. 44, 1400d, 93398. Gentle refluxing of 5.8 g. HC(SO3Ag)3 for 14
hrs. with 5 g. ELI in 30f ml. C666 yielded 2.4 g. nearly pure HC(SO2OEL)3
(I), fine needles, m. 75' (from C6H6-petr. ether). Br water added
to I in H2O causes an immediate white precipitate (III), which dissolves
rapidly
to a clear solution (III). Both II and III liberate iodine from KI. I in
absolute Etch reacts with Br immediately, less rapidly in Me2Co, and
slowly in
glacial AcOH, the addition of water to the glacial AcOH causes a
disappearance of the Br color. I in NaOH causes a slow decrease in the
concentration of NaOH, the anion of HC(SO2OH)3 being unstable in alkali.
A dilute
alkaline solution of HC(SO2OK)3 uses up alkali and, after acidification
the
solution, reduces NnO4- with the probable production of SO32-. The lat
step
in the hydrolysis of I was found to be catalyzed by H3O+. In water (no
mineral acid), the bimol. constant= 0.316 l./moles-min. and in 0.050 N
and
0.100 N HCl the pseudounimol. consts. = 0.0189 and 0.0336/min., resp.

The
hydrolysis measurements were carried out at 25° in a
glass-stoppered flask and periodic samples were titrated with KBr and Br
(Me orange) for the KCl catalyzed studies.

L20 ANSWER 26 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1952:26485 CAPLUS 1992:20485 CAPLUS
46:26485
46:4471d-e
Trisulfonylmethanes and methanetrisulfonates
Samen, Evald
Uppsala Univ., Swed.
Svensk Kem. Tid. (1951), 63, 31-41 DOCUMENT NUMBER: ORIGINAL REFERENCE NO .: TITLE: AUTHOR (S): CORPORATE SOURCE: SOURCE: DOCUMENT TYPE: Journal English UMAGE: English
873390-12-6, Methannetrisulfonic acid, triethyl ester
(preparation of)
873390-12-6 CAPLUS Methanetrisulfonic acid, triethyl ester (5CI) (CA INDEX NAME) OEt î CHcf. C.A. 45, 10190b. Review of work by S. and others, with complementary expts. Refluxing CH(SOZEL)3 with absolute alc. and CHCl3 produced no even with H2SO4 as catalyst. A trisulfone can be titrated with Br and (methyl orange indicator). CBr(SO2Et)3 refluxed with powdered Ag or K pellets in/C6H6 failed to produce C2(SO2Et)6, and a mixture of (CO2H)2,
PhSH, and anhydrous HCl produced no hexasulfide. Br added to CH(SO3Et)3 in H20 gives CBr(SO3Et)3 (II), needles, m. 60°-60.5°, and decomposing 110°. II liberates iodine from KI, but I does not.

L20 ANSWER 28 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1950:49172 CAPLUS
DOCUMENT NUMBER: 44:49172
ACH 19172
CORPORATE SOURCE: 5ame, Evald
Univ. of Uppsala, Swed.
SOURCE: 4Chemica Scandinavica (1950), 4, 397-8
COEN: ACHSE7; ISSN: 0904-213X
JOURNAL LANGUAGE: Unavailable
Unavailable
Unavailable
T 872793-71-0 CAPLUS
CN Methanetrisulfonic acid, trimethyl ester
(hydrolysis velocity of)
RN 872793-71-0 CAPLUS
CN Methanetrisulfonic acid, trimethyl ester

OSSOURCE: (CA INDEX NAME)

AB The hydrolysis of CH(SO3Me)3 (I) in aqueous HCl was studied by titration with

Br and an azo indicator (methyl orange). (It was assumed that only unhydrolyzed triester absorbed Br.) The reaction was found to be pseudounimol. I was prepared from MeI and CH(SO3Ag)3.

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L20 ANSWER 29 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1550:28491 CAPLUS DOCUMENT NUMBER: 44:28491 ORIGINAL REFERENCE NO.: 44:5552b-d
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44:305Zb-d Aluminum methionate Christian, John E.: Jenkins, Glenn L. Purdue Research Foundation Patent Unavailable

PATENT ASSIGNEE(S): DOCUMENT TYPE:

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	US 2504107		19500418	US 1946-701091	1946100
IT	855840~41-4, Metha (preparation of		lfonic acid,	aluminum salt	

855840-41-4 CAPLUS Methanetrisulfonic acid, aluminum salt (5CI) (CA INDEX NAME)

503° -03s-CH-SO3-

● Al 3+

Certain Al alkyl sulfonates are effective as antiperspirants, e.g. Al methionate (I), Al ethanedisulfonate, Al methanetrisulfonate, Al methanesulfonate, Al sulfoacetate, Al o-sulfobenzoate. General requirements are: the Al ion or its equivalentr one or more sulfo

alkyl group. The more sulfo groups, the greater is the astringency. I

prepared by adding a solution of Al2(SO4)3 to a solution of Ca

water, digesting the mixture for 10 min., and filtering it. The solution of I

is concentrated, and EtOH is added until I crystallizes out. I is hygroscopic, soluble in water, does not crystalline out of astringent creams. The

other Al salts are prepared by treating the respective Ba compds. with Al2(504)3 solution These antiperspirants are harmless to skin and fabrics.

L20 ANSWER 31 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1949:27346 CAPLUS
DOCUMENT NUMBER: 43:27346
A3:27346
A3:27

LANGUAGE: Patent Unavailable FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE FR 867066 19410926 FR
54322-33-7, Methanetrisulfonic acid
(catalyst of kgO, HgSO4 and, in C2H2 reaction with AcOH)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME) 19400224

SO3H

HO3S-CH-SO3H

With a mixture of sulfonic acids containing CH-(SO3H)3 36, HgSO4 21, and

4
as catalyst, AcOH 5600 g. and C2H2 give CH2CHOAc or MeCH(OAc)2. The sulfonic acid mixture is prepared by adding 631 H2SO4 1070 to Ac2O 400 g. slowly so that the temperature does not rise above 115', letting stand 3 hrs. at 120', adding glacial AcOH 500 g. with the temperature at 120' another 3 hrs., and then more glacial AcOH 300 g., with stirring 0.5 hr. The mixture contains 15% CH (SO3H)3.

ACCESSION NUMBER: DOCUMENT NUMBER: ORIGINAL REFERENCE NO.:

AUTHOR (S):

ANSWER 30 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
SSION NUMBER: 1950:7342 CAPLUS
MENT NUMBER: 44:7342
HINAL REFERENCE NO.: 44:1400d-f
E: 44:1400d-f
Methyl methanetrisulfonate
Samen, Evald
Arkiv foer Kemi (1949), 1, 231-3
COEN: ARKEAD, ISSN: 0365-6128
MENT TYPE: Journal
BURGE: English

DOCUMENT TYPE:

DOCUMENT 11P2.

English
LINGUAGE: English
Lingua

AB cf. C.A. 37, 5015.8. Tri-Me methanetrisulfonate (I) was prepared for the first time by refluxing 11.5 g. dry CH(5020Ag)3 (cf. C.A. 25, 915) 8 hrs. with 9 g. MeI in 50 ml. C6H6; the filtrate from the precipitated AgI deposited

itted
1.8 g. I which separated from boiling C6H6 in long fine needles, m.
110-11°. I possesses strong acid properties and is sparingly soluble
in C6H6, Et2O, and CHCl3, but more soluble in H2O. I (68.09 mg.) in 10

boiling H2O was hydrolyzed in 1 min.; a 0.01 M solution in H2O was half hydrolyzed in 20 hrs. at room temperature I $(0.6\,\mathrm{g}_{\odot})$ in 10 ml. H2O at 0° reacted immediately with an equivalent amount of Br in cold H2O, forming

0.4 g.
C4H9O9BrS3 (II), m. 140°. II liberated iodine from KI instantly and quantitatively. Br in AcOH reacted slowly with I in the same

solvent. Addition of H2O caused the Br color to vanish immediately.

L20 ANSMER 32 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1947:11308 CAPLUS
CORGINAL NUMBER: 41:11308
CAPLUS
CORGINAL REFERENCE NO: 11:203e-g
Diameter changes of gelatinized coacervate drops of the complex coacervate gelatin-gum arabic, resulting from a change in pH of, or from the addition of neutral salts to, the surrounding medium. I
CORPORATE SOURCE: CORPORATE SOURCE: Bungenberg: Landsmeer, J. M. F.
Univ., Leiden
CODEN: RTCPB4; ISSN: 0370-7539
DOCUMENT TYPE:

DOCUMENT TYPE: Journal

DOCUMENT TIPE.

LINGUAGE:

Style="background-color: blue;">DOCUMENT TIPE.

DOCUMENT TIPE.

DOC

SO3H нозя-сн- возн

●з к

Diameter changes of coacervate drops (100-200 µ) reached equilibrium quickly (5-10 min.) in media of a given pH or salt concentration. The changes

followed microscopically at salt concns. low enough (10-40 milliequiv.

1.) to allow reversibility in the swelling and shrinking. The diameter min. at the pH (3.7) where the H2O content is a min. Reversible swelling due to addition of a salt (KCl, CaCl2, LaCl3, K2SO4, or K3CH(SO3)3) is

least for the 1-1 valence type and increases as the salt deviates from this valence type (double valence rule). L20 ANSWER 33 OF 37 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1944:16250 CAPLUS DOCUMENT NUMBER: 38:16250

ACCESSION NUMBER: DOCUMENT NUMBER: ORIGINAL REFERENCE NO.:

38:2347f-g Reaction of oleum with AcOH or Ac2O Cockerille, Frank O. E. I. du Pont de Nemours & Co. Patent TITLE: INVENTOR(S):

PATENT ASSIGNEE(S): DOCUMENT TYPE: LANGUAGE:

Unavailable

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

APPLICATION NO. PATENT NO. KIND DATE DATE US 2333701 19431109 54322-33-7, Methanetrisulfonic acid 19431109 US 1940-365947 19401116 IT

(preparation of) 54322-33-7 CAPLUS

Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

șo3H

нозв-сн- возн

Methanetrisulfonic acid is prepared by a process involving adding HOAc or Ac2O to oleum, in a proportion of 4 to 7 parts of oleum to each part of the HOAc or Ac2O, the temperature being allowed to rise gradually to 65-85 during the addition of 50-751. and not above 90° during the addition of the remainder, of the Ac2O or HOAc, and completing the reaction by holding the mixture at 90-100°.

ANSWER 34 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) (010):(111) = 54° 5'; (001):(111) = 57° 30'; (001):- (102) = 25° 25'; (010):(.hivin:10) = 43° 50'; plane of the optical axes (100); the sharp bissectrix is (010); p > v; the crystals are optically negative. The normal T1 salt is anhyd. Normal Ba salt (9H2O); at 25°. 100 g, water dissolves 0.845 g, anhyd. salt; crystallographic properties: thombic system; a:b:c = 0.841:10.564; observed forms (212), (010), (110), (111) and (120); angles:(010):(110) = 50° 11'; (010):(121) = 36° 31' 30'; plane of the optical axes (010). Normal NH4 salt, anhyd.; crystallographic properties:rhombic system;

a:b:c c = 0.9573:1:1.7022; observed forms (001), (011), (101), (110) and (211); angles:(001):- (011) = 59° 34°; (100):(101) = 29° 21°; (100):(110) = 43° 44°; (100):(211) = 29° 16°. PC15 with the free acid does not give the trieulfonyl chloride, SO2 being evolved; therefore, derivs. could not be prepd.

L20 ANSWER 34 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1932:23284 CAPLUS
DOCUMENT NUMBER: 26:23284
ORIGINAL REFERENCE NO: 26:2413g-1,2414a-d
TITLE: The chlorination of methanetrisulfonic acid
AUTHOR(S): Backer, H. J.
SOURCE: Recueil des Travaux Chimiques des Pays-Bas et de la
Belqique (1932), 51, 374-80
CODEN: RTCPB4; ISSN: 0370-7539

Journal Unavailable LANGUAGE: Unavailable IT 54322-33-7, Methanetrisulfonic acid

(chlorination of)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H

DOCUMENT TYPE:

HO3S-CH-SO3H

cf. C. A. 25, 75. The chlorination of CH(SO3H)3 may be carried out by heating 7.8 g. of the tri-K salt and 0.1 g. I in 60 cc. water containing

Cl in a sealed tube at 200° for 10 hrs. After evaporating the contents of the tube and recrystg. the residue 6 times from its weight of water,

Chloromethanetrisulfonate was obtained in a pure state in 70% yield; it does not contain H20; at 25° 100 g. water dissolves 3.38 g.; crystallographic properties: monoclinic system; a:b:c = 0.8040:1:0.7659; β = 87° 52°; observed forms (.hivin:111), (010, (111), (101) and (011); angles: (010):(011) \$2° 34°; (010):(111) = 61° 27°; (100):(.hivin:101) = 42° 36°; (100):(101) = 44° 33°; (010):(.hivin:111) = 60° 38°. For the preparation of the free acid and other salts the tristrychnine salt, which crystallizes without H20, was used. The free acid (4.5 H20) m. 160-5° but 180-2° in the presence of F205; at 25°, 100 g. water dissolves 254.7 g. of the anhydrous acid. Normal Li salt (8 H20): 100 g. water dissolves 109.9 g.

the anhydrous salt; crystalline properties: rhombic system; a:b:c = 0.933:1:0.708;

0.933:1:0.708;

observed forms (110), (011) and (010); angles: (110):(010) = 46°
59'; (1.havin.11):(100) = 52° 43'; (010):(011) = 54° 50';
(110): (011) = 66° 51'. The normal Na salt does not contain H2O on crystallization from a saturated hot solution but 3 H2O on evaporating a solution at room temperature;
at 25°, 100 g. water dissolves 21.8 g. of the trihydrate. Normal Rb salt (1H2O): 100 g. water dissolves at 25°, 3.34 g. of the anhydrous salt; crystalline properties: rhombic system; a:b:c = 0.9827:1:0.5437;
observed forms (1011, 4010)

27:1:0.5437;
observed forms (101), (010), (111), (221), (100), (041), (001) and (210), sometimes (110); angles: (001):(111) = 37* 48'; (100):(110) = 45* 30'; (001):(101) = 25* 1'; (001):(221) = 57* 20'; (001):(041) = 55* 23'; (100):(210) = 26* 12'; (101):(111) = 25* 20'. The normal Cs salt (1 H20) consists of rhombic plates; at 25*, 100 g. water dissolves 5.49 g. of the crystallized salt; crystallographic properties: rhombic system; a:b:c = 0.9646:1:1.0845; observed forms (100), (001), (010), (111), (102), (101), (210) and (112); angles:(100):(101) = 41* 39'; (100):(210) = 25* 45'; (100):(112) = 63* 47'; (100):(111) = 53* 0';

L20 ANSWER 35 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1932:8729 CAPLUS
DOCUMENT NUMBER: 26:8729
ORIGINAR REFERENCE NO: 26:9829-1
TITLE: The salts of methanetrisulfonic acid
AUTHOR(s): Backer, H. J.; Terpstra, P.
SOURCE: Recueil des Travaux Chimiques des Pays-Bas et de la
Belqique (1931), 50, 1069-77
CODEN: RTCPB4; ISSN: 0370-7539

Journal Unavailable

CODEN: RTCPB4; ISSN: 0370-7339
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable
IT 54322-33-7, Methanetrisulfonic acid
(and salts)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

ŞO3H

HO35-CH-SO3H

AB The normal K, Ag and Tl salts of methane-trisulfonic acid have been described (cf. Backer, C. A. 25, 915); the present paper deals with other salts, which were prepared in the usual way. Methanetrisulfonic acid contains 3H20, m. 162-162.5; 100 g. of water at 25° dissolves 221 g. of the anhydrous acid and 510 g. of the crystallized compound

Detailed crystallographic data are given for the normal NH4 salt (contains

0.5 H20; crystals are of the monoclinic system), the normal Li salt (4 H20; monoclinic system), normal Na salt (3 H20; very thin monoclinic plates), normal Rb salt (1 H20; rhombic system), tsphenoidal class), normal Cs salt (1 H20; monoclinic system), normal Ag salt (1 H20; trielinic system), normal Ca salt (12 H20; rhombic system), normal Ba salt

(9 H2O; monoclinic crystals), normal La salt (6 H2O; monoclinic

(9 H2O; monoclinic crystals), normal La salt (6 H2O; monoclinic crystals), and normal cinchonine salt (8 H2O; rhombic, pseudotetragonal crystals). The mono-K dibrucine salt was obtained from the tri-K salt and an equivalent amount of brucine acetate; it contains 9 H2O. The K and Rb salts are isomorphous and give mixed crystals. Sometimes small anhydrous triclinic crystals of the Cs salt were obtained and a triclinic dihydrate was prepared

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L20 ANSWER 36 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1931:8598 CAPLUS
DOCUMENT NUMBER: 25:8598
ORIGINAL REFERENCE NO.: 25:915a-1
                                                                              23:913a-1
Methanetrisulfonic acid
Backer, H. J.: Klaassens, K. H.
Recueil des Travaux Kchiniques des Pays-Bas et de la
Belgique (1931), 49, 1107-17
CODEN: RTCP84; 158N. 0370-7539
  AUTHOR (S):
 SOURCE:
 DOCUMENT TYPE:
                                                                                Journal
                                                                               Unavailable
  LANGUAGE: Unavailable
IT 54322-33-7, Methanetrisulfonic acid
               (and salts)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                   SO3H
HORS-CH-SOLH
                CH(SO3H)3 has already been prepared: (1) by Theilkuhl (Ann. 147, 134
                  on heating Ca Me sulfate with H2S2O7; (2) by Bagnall (J. Chemical
(von Pechmann, Ber. 28, 2382(1895); Fantl and Fisch, C. A. 24, 1841); (6) by sulfonation of CH2(SO3H)2. All these methods, except 1, are discussed in the present paper. Method 6: On heating CH2(SO3H)2 with SO3 (2.5 mols.) during 5 hrs. at 170°, CH(SO3H)3 may be obtained as the K salt with 10% yield. Method 2: sulfonation of PhNHAc 0.5 mol. AcNNPh is introduced slowly into 600 g. fuming H2SO4 containing 35% SO3, the ture being heated 3 hrs. at 130° with mech. stirring; yield of the tri-K salt 57%; on carrying out the same reaction with succinanilide, a small yield of the same compound was obtained. Method 3: 0.1 mol. HSC(SO3K)3, in 1
water is oxidized at room temperature by a current of Cl with the gradual addition of 80 g. KHCO3; yield 90%. Method 4: O2NCH(SO3H)2 was prepared according to 80 g. KHCO3; yield 90%. Method 4: O2NCH3, gradually to 450 g. K2SO3 in 900 cc. water, heated to 75°, and keeping the temperature at 80°; yield 40% of the anhydrous di-K salt, which dissolves in water at 25° to the extent of 1.16%. The normal strychnine salt of O2NCH(SO3H)2 crystallizes with 3.5 H2O, the free acid itself with 2 H2O while the normal Na and Tl salte crystallize without H2O; both the latter salts are easily soluble in water. On heating the K salt with K2SO3 in a sealed tube
                at 140°, CH(SO3K)3 is obtained in 67% yield. Method 5: N2C(SO3K)2, prepared according to von Pechmann (loc. cit.), may be converted into CH (SO3M)3 by the action of KHSO3 at 60^\circ and finally on the water bath (yield 62%), decomposing the compound with water or introducing it into
 dilute
L20 ANSWER 37 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1930:16947 CAPLUS
DOCUMENT NUMBER: 24:16847
COPYRIGHAIR REFERENCE NO.: 24:1841a-b
ROTHOR(S): 24:1841a-b
ROTHOR(S): Fantl, Paul; Fisch, Julius
SOURCE: Journal there Praktische Chemie (Leipzig) (1930), 124, 159-62
CODEN: JOURNAL TYPE: JOURNAL LANGUAGE: Unavailable
LANGUAGE: Unavailable Unavailable
(preparation of)
RN 55110-91-3 CAPLUS
CN Methanetrisulfonic acid, hydroxy-, tripotassium salt
(proposation of)
RN 55110-91-3 CAPLUS
CN Methanetrisulfonic acid, tripotassium salt (3CI, 9CI) (CA INDEX NAME)
 нозя-сн- sозн
             ●3 K
             54322-33-7, Methanetrisulfonic acid
                (salts)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                   șo<sub>3</sub>H
  нозя- cн- sозн
                N2C(SO3K)2 and KHSO3 give CH(SO3K)3 and not HOC(SO3K)3 (cf. v. Pechmann, Ber. 28, 2374(1898)). The K, Ba and Ag salts were prepared and analyzed
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L20 ANSWER 36 OF 37 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) HCl, the latter reaction giving a 401 yield. On introducing the diazo compd. into concol. HCl at 0°, ClCH(SO3H)2 is obtained. The following salts of CH(SO3H)3 and the acid itself were prepd. in the usual way: The free acid with 3.5 H2O, m. 156°; the tri-K salt, rhombic prisms with 1 H2O; 100g. of water dissolve at 25°,1.202 g. of the anhyd. salt and 1.261 g. of the crystd. salt. Crystallographic properties

(P. TEMPSTRA): rhombic system, bisphenoidic class D2; a: b: c = 0.9977:11:1.3604. Forms: a = (1001; b = (0101; c = (001); p = (101); m = (1101); o = (111); f = (1.hivin.11); q = (011); s = (121); angles: a: m = 44° 56'; c: p = 52° 38'; a: * = 51° 28'; b: s = 51° 34'; c: q = 52° 36'; o: s = 19° 22'. The crystals obtained by evapn. at ordinary temp. have a larger dimension in the direction of the b-axis. The crystals show a marked piezoelectricity; the plane of the optical axes is (001); the a-axis is the sharp bissectrix; p<v. For Na light 2c 83° 1'; c = 1.513; β = 1.5253; y = 1.5270; 2v = 51° 45'. The tri-T1 salt also contains H2O; crystallographic properties (P. TEMPSTRA): rhombic system, bisphenoidic class: a: b: c = 0.9971:1:-13009; forms: m = (110); p = (101); q = (011); w = (1.hivin.11); " = (111); the crystals possess either the form or the form which have never been found to occur in the same crystal. Angles: m: m = 90° 10'; p: p = 105°4'; q: q = 104°54'; a: w = 123°0'; p: w = 38°20'; q: q = 38°29'. The plane of the optical axes is (001); the sharp bissectrix (010). By means of a prism formed by (1.hivin.11) and (1.hivin.11) the refractive index has been detd. to be 1.739, 1.743, 1.768 for x = 578, 546, 436, resp. The crystals are piezoelectrical. The tri-Ag salt contains 1 H2O; the Ba salt, 9 H2O; the latter is difficulty sol. in water, only to the extent of 0.1%. On mixing without precautions the solns of the K salt and BaCl2, a double salt of Ba and K is obtained, ChO993KBB.3H2O, which,

several recrystns. from a large amount of water, gives the pure Ba salt. The Ca salt contains 12 H2O and the La salt 6 H2O (cf. Backer and Klaassens, C. A. 24, 4729).

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http://www.cas.org/ONLINE/UG/regprops.html

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chain nodes:
1 2 3 4 5
chain bonds:
1-4 2-4 3-4 4-5
exact bonds:
1-4 2-4 3-4 4-5

Match level:
1:CLASS 2:CLASS 3:CLASS 4:CLASS 5:CLASS

L21 STRUCTURE UPLOADED

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FULL SCREEN SEARCH COMPLETED - 99 TO ITERATE

100.0% PROCESSED

99 ITERATIONS

9 ANSWERS

SEARCH TIME: 00.00.01

L22

9 SEA SSS FUL L21

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TOTAL

FULL ESTIMATED COST

ENTRY 166.94 SESSION 921.18

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

SINCE FILE

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=> s 122

L23 30 L22

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L23 ANSWER 1 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
                                                        (Continued)
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L23 ANSWER 1 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 2005:426557 CAPLUS DOCUMENT NUMBER: 142:463900
                                                                                             Acylation process for the preparation of 2,3,5-trimethylhydroquinone diacylates
 TITLE:
                                                                                            of methanetrisulfonic acid catalyst
Aquino, Fabrice; Bonrath, Werner: Pa
DSN IP Assets B.V., Neth.
PCT Int. Appl., 10 pp.
CODEN: PIXXU2
Patent
INVENTOR(S):
  PATENT ASSIGNEE (S):
 SOURCE:
DOCUMENT TYPE:
                                                                                              English
 FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
              PATENT NO. KIND DATE APPUICATION NO. DATE

WO 2005044775 Al 20050519 WO 2004-EP12058 20041026
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GG, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, NA, MD, MG, MK, MN, MM, MZ, NA, NI, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TM, TT, TZ, YA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
RW: BW, GH, GM, KE, LS, MM, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SP 1680392 R: AT, BE, CH, DE, DATE
SN. TD, TG

EP 1680392

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,

IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK

PRIORITY APPLIN. INFO:

EP 2003-25513

A 20031107
                                                                                                                                                                    WO 2004-EP12058
                                                                                                                                                                                                                                               W 20041026
OTHER SOURCE(S):

CASREACT 142:463900

IT 54322-33-7, Methanetrisulfonic acid
RL: CAT (Catalyst úse): USES (Uses)

(acylation process for preparation of 2,3,5-trimethylhydroquinone diacylates

in presence of methanetrisulfonic acid catalyst)

RN 54322-33-7 Capfus

CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                      SO3H
нозя-сн-sозн
AB 2,3,5-Trimethyl-1,4-hydroquinone diacylates (e.g., 2,3,5-trimethyl-1,4-benzoquinone diacetate), useful as intermediates in the preparation of a-tocopherol (no data), are obtained in high yield and selectivity by feating 3,5-trimethyl-1,4-benzoquinone with an acylating agent (e.g., acetic anhydride) in the presence of catalytic quantities of methanetrisulfonic acid.

REFERENCE COUNT: 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THI
                                                                                                                  THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE
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L23 ANSWER 2 OF 30
ACCESSION NUMBER:
DOCUMENT NUMBER:
111:395687
Process for the manufacture of tocyl and tocopheryl acylates
Bonrath, Werner; Haas, Alois; Hoppmann, Simone;
Netscher, Thomas; Pauling, Horst
DSM IP Assets B.V., Neth.
PCT Int. Appl., 15 pp.
CODEN: PIXXD2
DOCUMENT TYPE:
Patent

COPPRIGHT 2006 ACS on STN
ACCESSION ACCESSION STN
ACCESSION ACCESSION STN
ACCESSION ACCESSION STN
ACCESSION ACCESSION
        DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                                                                                   Patent
English
PATENT NO.
                                                                                                                                                                                                                                   KIND
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      DATE
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ES, FI, GB, GD,
KP, KR, KZ, LC,
MC, MZ, NA, NI,
SG, SK, SL, SY,
YU, ZA, ZM, ZW
ZM, ZW, AM, AZ,
CZ, DE, DK, EE,
PT, RO, SE, SI,
ML, MR, NE, SN,
        OTHER SOURCE(S): CASREACT 141:395687; MARPAT 141:395687

IT 54322-33-7, Methanetrisulfonic acid
RI: CAT (Catalyst use) / USES (Uses)
(process for manufacture of tocol and tocopherol acylates using
perfluoroalkylsulfonyl catalysts)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                           503H
           нозя-сн- возн
           GI
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L23 ANSWER 2 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) or a tocopheryl acylate I [R = acyl; R5 = R7 = R8 = Me, R5 = H, R7 = R8 = Me, etc.] comprised reacting a corresponding tocol or tocopherol with an acylating agent in the presence of a catalyst of the general formula HCRIR2R3 [wherein R1, R2 and R3 each signify the sulfo group, or R1, R2 and R3 each signify a perfluoroalkylaulfonyl group whereby at least two of R1, R2 and R3 are identical such perfluoroalkylaulfonyl group and R2 and R3 each signify an identical perfluoroalkylsulfonyl group. The main com. form of vitamin E, being (all-rac) accompletely acetate I [R = acetyl; R5 = R7 = R8 = Me], can be manufd. by acylation of (all-rac) actoopherol according to this process.

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE
       FORMAT
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A process for the manufacture of tocyl acylate I (R = acyl; R1 = R2 = R3

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L23 ANSWER 3 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 2004:823008 CAPLUS DOCUMENT NUMBER: 141:334863
                                                                                             141:334663
Crosslinked polyoxyalkylene-polysiloxanes for use as nonaqueous salt-type electrolytes for lithium secondary batteries
Barrandon, Georges; George, Catherine; Vergelati, Caroll; Giraud, Yves
Rhodia Chimie, Fr.
Fr. Demande, 25 pp.
CODEN: FRXXBL
  TITLE:
INVENTOR(S):
 PATENT ASSIGNEE(S):
 DOCUMENT TYPE:
                                                                                               Patent
   LANGUAGE:
  FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                    PATENT NO.
                                                                                              KIND
                                                                                                                    DATE
                                                                                                                                                                    APPLICATION NO.
                                                                                                                                                                                                                                                           DATE
                    FR 2853321
                                                                                                                        20041008
20050506
20041021
                                                                                                                                                                                                                                                           20030403
                                                                                                                                                                    FR 2003-4153
                                                                                                 A1
B1
                     FR 2853321
                    WO 2004090037
                                                                                                  Al
                                                                                                                                                                     WO 2004-FR708
                                                                                                                                                                                                                                                            20040323
                  MO 2004-9708 20040323

N: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
GE, GH, GM, HR, HU, ID, IL, NI, IS, JF, KE, KG, KF, KR, KZ, LC,
LK, LR, LS, LT, LU, LV, MA, MD, MG, KK, MN, MW, KK, AZ, NA, NI,
NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
TJ, TM, TN, TT, TT, TZ, UA, UG, US, UZ, VC, VN, VU, AZ, AM, NZ
RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI,
SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
TD, TG

EP 1608705 A1 20051228 EP 2004-742318 20040323
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
                                              TD, TG A1 20051228 EP 2004-742318 20040323
AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK
91054 A 20060614 CN 2004-80013072 20040323
PLN. INFO:: FR 2003-4153 A 20030403
                     CN 1788054
 PRIORITY APPLN. INFO.:
                                                                                                                                                                                                                                                W 20040323
                                                                                                                                                                     WO 2004-ER708
OTHER SOURCE(S): MARPAT 141:334863
IT 54322-33-7DP, Methanetrisulfonic acid, derivs., salts
RL: DEV (Device component use): SPN (synthetic preparation); TEM
(Technical or engineered material use): PREP (Preparation); USES (Uses)
(battery electrolytes containing; crosslinked
polyoxyalkylene-polysiloxanes
polyoxyalkylene-polysiloxanes
batteries)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                       ŞO3H
 HO3S-CH-SO3H
                    Crosslinked polymeric electrolytes for lithium secondary batteries
                     st of: (1) a first poly(hydrogen organic siloxane) with \geq2 Si-H bonds per mol., (2) a second polysiloxane containing \geq2 Si-OH bonds per mol., (3)
  L23 ANSWER 4 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 2004:823007 CAPLUS
DOCUMENT NUMBER: 41:334862
Fintochemical crosslinked epoxy-polyoxyalkylene-polysiloxanes for use as nonaqueous salt-type electrolytes for lithium secondary batteries
INVENTOR(S): Gambut, Lucile: Vergelati, Caroll: Sanchez, Jean
                                                                                              Alloin, Fannie
Rhodia Chimie, Fr.; Rhone Poulenc Chimie
Fr. Demande, 30 pp.
CODEN: FRXXBL
Patent
French
  PATENT ASSIGNEE(S):
SOURCE:
   DOCUMENT TYPE:
   LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                      APPLICATION NO.
PATENT NO. KIND DATE APPLICATION NO. DATE

FR 2853320 A1 20041008 FR 2003-4150 20030403
FR 2853320 B1 20050506
W0 2004091033 A2 20041021 W0 2004-FR707 20040323
W0 2004091033 A3 20050714
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, UI, DI, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MM, MM, MX, MZ, NA, NI, NO, NZ, CM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TM, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
RW: BW, GH, GM, KZ, LS, MM, MZ, BS, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, SK, TR, BE, BJ, CF, CG, CI, CM, GA, GM, GM, MM, MM, MM, MR, NN, MR, NR, NN, TD, TD, TG

EP 1609205 A2 20051228 EP 2004-742317 20040323
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IF, SG, LT, LT, LV, FI, RO, GN, CY, AL, TR, BG, CZ, EE, ML, PL, SK
CN 1784806 A 20060607 CN 2004-80012563 200400323
PRIORITY APPLN. INFO::
                      PATENT NO.
                                                                                                                                                                                                                                                             DATE
                                                                                                                                                                       WO 2004-FR707
                                                                                                                                                                                                                                                W 20040323
  OTHER SOURCE(S): MARPAT 141:334862

IT 54322-33-7DP, Methanetrisulfonic acid, derivs., salts
RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use): PREP (Preparation); USES (Uses)
(battery electrolytes containing: photochem. crosslinked epoxy-polyoxyalkylene-polysioxanes for use as nonaq. salt-type electrolytes for lithium secondary batteries)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                         SO3H
   HO25-CH-SO2H
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Crosslinked polymeric electrolytes for lithium secondary batteries are prepared from a polyorganosiloxane grafted with polyoxyalkylene and functions, followed by radiochem. or photochem. (e.g., UV-induced)

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L23 ANSWER 3 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) a dehydrogenation-condensation catalyst, and (4) ≥1 salt electrolyte. The polyoxyalkylene ether functions are derived from polyoxyethylene, polyoxypropylene, or their mono-Me ethers. The dehydrogenation-condensation catalysts are typically metal complexes
                       on Pt, B, Rh, Pd, Sn, or Ir, preferably Karstedt (hydrosilylation) catalysts of formula IrCl(C:O)(PPh3)2. Suitable salt electrolytes
catalysts of formula five(co, theorem) catalysts of formula five(co, theorem) catalysts of formula five(co, theorem) catalysts and Lin(C2F5SO2)2 in a non-aq. electrolyte solvent, as well as other cations (e.g., transition metal cations, selected from Mn, Fe, Co, Ni, Cu, Zn, Ca, and Aq). Addnl. ions include ammonium, amidinium, guanidinium cations, halides, ClO4-, SCN-, Bf4-, NO3-, Asf6-, Pf6-, RO3- (R = steary), Cf3, octy), dodecylphenyl, and C1-6-perfluoroalkyl and -perfluoroaryl), (R5SO2)2N-, and (R4SO2)(R5SO2) (R6SO2)C- (R4-6 = C1-6-perfluoroalkyl and -perfluoroaryl).
 -perfluoroaryl).
REFERENCE COUNT:
                                                                                                                                                THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE
FORMAT
 L23 ANSWER 4 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) crosslinking in the presence of a photosensitizer. The crosslinked epoxy-polyoxyalkylene-polysiloxanes have the general formula RixRyR3z5j0(4-x-z)/2, in which R1-3 are C1-12-alkyl, C5-10-cycloalkyl, C6-18-aryl, aralkyl, or -OR4 (R4 = H or C1-15-alkyl; and x + y + z = 1-3).
                       The product polysiloxanes (typically prepd. from Si-H bond-contg. polysiloxanes by hydrosilylation grafting with allyl-terminated polyoxyalkylenes and epoxides) are then combined with org. salts to form the battery electrolytes. Suitable salt electrolytes include Liclo4, LiBF4, LiAsF6, CF3SO3Li, LiN(CF3SO2)2, and LiN(CFF3SO2)2, other cations (e.g., transition metal cations); ammonium, amidinium, and guantidinium salts; org. sulfonates, imidodisulfonates, methanetrisulfonates, organoboron complex salts, ferricenium salts, and other onlum salts (esp. iodonium and sulfonium compds.).
RENCE COUNT:

4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE
  REFERENCE COUNT:
  FORMAT
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L23 ANSWER 5 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 2004:453200 CAPLUS DOCUMENT NUMBER: 141:23730 MANUFACTURE OF CAPPELS
                                                                                                                                                                                                                                                  126:211808
Investigations on the acid-strength of alkanepolysulfonic acids
Jueschke, Ralf; Sartori, Peter
Fachbereich 6 Anorganische Chem., Gerhard-Mercator
Univ., Duisburg, D-47048, Germany
Zeitschrift fuer Naturforschung, B: Chemical Sciences
(1996), 51(12), 1691-1700
CODEN: ZNBSEN; ISSN: 0932-0776
Verlag der Zeitschrift fuer Naturforschung
Journal
German
                                                   141:23750

Manufacture of a-tocopherol from the reaction of trimethylhydroquinone with isophytol or phytol in the presence of methane trisulfonate
Bonrath, Werner; Hoppmann, Simone; Haas, Alois; Netscher, Thomas; Pauling, Horst
DSM IP Assets B.V., Neth.
PCT Int. Appl., 13 pp.
CODEN: PIXXD2
Patent
                                                                                                                                                                                              DOCUMENT NUMBER:
TITLE:
                                                                                                                                                                                              AUTHOR(S):
CORPORATE SOURCE:
INVENTOR (S):
 PATENT ASSIGNEE(S):
                                                                                                                                                                                              SOURCE:
 DOCUMENT TYPE:
                                                    Patent
                                                                                                                                                                                               PUBLISHER:
                                                                                                                                                                                              DOCUMENT TYPE:
                                                    English
                                                                                                                                                                                                        AUGE: German
54322-33-7, Methanetrisulfonic acid
RL: PRP (Properties)
(preparation of silyl alkanepolysulfonates and estimation of acidity
 FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                                                                                                               LANGUAGE:
           PATENT NO.
                                                    KIND
                                                                 DATE
                                                                                            APPLICATION NO.
                                                                                                                                             DATE
                correlation with 29Si-NMR shift)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
           WO 2004046127
                                                                                                                                                                                                          SO3H
                                                                                                                                                                                              нозя-сн- возн
                                                                                                                                                                                                         55110-91-3
RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation of silyl alkanepolyaulfonates and estimation of acidity
                                                                                                                                                                                              ΙT
           AU 2003270295
CN 1701065
US 2006020139
                                                                                                                                                                                              bу
                                                                                                                                                                                                               correlation with 29Si-NMR shift)
 PRIORITY APPLN. INFO.:
                                                                                                                                                                                                         55110-91-3 CAPLUS
Methanetrisulfonic acid, tripotassium salt (3CI, 9CI) (CA INDEX NAME)
                                                                                                                                    W 20030930
                                                                                            WO 2003-EP10837
OTHER SOURCE(S): CASREACT 141:23750
IT 54322-33-7, Methanetrisulfonic acid
RL: CAT (Catalyst use): USES (Uses)
(manufacture of a-tocopherol from the reaction of
trimethylhydroquinone with isophytol or phytol in the presence of
methane trisulfonate)
RN 54322-33-7 CAPUSONATE)
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                                          $O3H
                                                                                                                                                                                              нозя-сн- возн
                                                                                                                                                                                                      ●3 к
                                                                                                                                                                                                         75533-77-6P 187610-57-7P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
            503H
                                                                                                                                                                                                          (Reactant or reagent)
  (preparation of silyl alkanepolysulfonates and estimation of acidity
 нозя-сн- возн
                                                                                                                                                                                              by
           (all-rac)-\alpha-tocopherol is prepared by the acid-catalyzed reaction of trimethylhydroquinone with isophytol or phytol in the presence of metrisulfonate as the catalyst in an organic solvent.
                                                                                                                                                                                                                correlation with 29Si-NMR shift)
                                                                                                                                                                                                         75533-77-6 CAPLUS
Methanetrisulfonic acid, barium salt (2:3) (9CI) (CA INDEX NAME)
                                                                                                                                                                                               L23 ANSWER 7 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1994:630311 CAPLUS DOCUMENT NUMBER: 121:230311
 L23 ANSWER 6 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
                                                                                                                                                                                                                                                    121:230311
Improvement of the synthesis of the tripotassium methanetrisulfonate monohydrate
Sartori, Peter; Jusechke, Ralf
Fachbereich 6 Anorg. Chem., Univ.-CH, Duisburg,
             503H
                                                                                                                                                                                               TITLE:
 HO3S-CH-SO3H
                                                                                                                                                                                               AUTHOR (S) :
                                                                                                                                                                                              CORPORATE SOURCE:

Fachbereich 6 Anorg. Chem., Univ.-CH, Duisburg,
Germany

SOURCE:

Journal tuer Praktische Chemie/Chemiker-Zeitung
(1994), 336(4), 373-4
(CODEN: JPCCEM; ISSN: 0941-1216

DOCUMENT TYPE:
JOURNAL

CASREACT 121:23031

IT $4322-33-7, Methanetrisulfonic acid
RL: FMU (Formation, unclassified); RCT (Reactant); FORM (Formation, nonpreparative); RACT (Reactant or reagent)
[Improvement of the synthesis of the tripotassium methanetrisulfonate monohydrate]
RN $4322-33-7 CAPLUS

CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                               CORPORATE SOURCE:
     ●3/2 Ba
           187610-57-7 CAPLUS Methanetrisulfonic acid, trisilver(1+) salt (9CI) (CA INDEX NAME)
             SO3H
 нозя-сн-возн
    ●3 Ag(I)
                                                                                                                                                                                                           SO3H
            The acidity of alkanepolysulfonic acids was measured using the
 correlation between the 295i-NMR shift of the corresponding Me3Si ester of an acid
                                                                                                                                                                                               нозв-сн- возн
           the pKS1 of the acid. The alkanepolysulfonates RCH(SO3SiMe3)2, RZC(SO3SiMe3)2, RC(SO3SiMe3)3, and Me3SiO3S(CR12)nSO3SiMe3 (R = H, F, Cl, Br, I; Rl = H, F; n = 1-5) were prepared and studied.
                                                                                                                                                                                                          73224-87-0P
                                                                                                                                                                                                         73224-87-0P
RI: SPN (Synthetic preparation); PREP (Preparation)
(improvement of the synthesis of the tripotassium methanetrisulfonate
monohydrate)
73224-87-0 CAPLUS
Methanetrisulfonic acid, tripotassium salt, monohydrate (9CI) (CA INDEX
NAME)
                                                                                                                                                                                                           SO3H
                                                                                                                                                                                               HO35-CH-SO3H
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●3 K

● H₂O

L23 ANSWER 6 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1997:69364 CAPLUS DOCUMENT NUMBER: 126:211808

The Bagnall-Backer method for the preparation of the title compound [via decarboxylation of acetanilide in SO3/H2SO4 to give [MO38]2C6H3MH2 and HC(SO3H3) (I) then CaO and K2CO3 to give HC(SO3H3)-H2O [III] is improved by the decarboxylation of acetone in SO3/H2SO4 to give I which with KOH gives 499 II.

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Robinson, E. A.
Erindale Coll., Univ. Toronto, Mississauga, ON, L5L
                                                                                                                                                                                                                                                                                                       Jones, Glenn C.; Nottingham, William D.; Raynolds,
Peter W.
Eastman Kodak Co., USA
                                                                THEOCHEM (1989), 55, 29-41
CODEN: THEODJ; ISSN: 0166-1280
                                                                                                                                                                                                                                       PATENT ASSIGNEE (S):
SOURCE:
                                                                                                                                                                                                                                                                                                       U.S., 4 pp.
CODEN: USXXAM
 DOCUMENT TYPE:
                                                                                                                                                                                                                                       DOCUMENT TYPE:
LANGUAGE:
                                                                                                                                                                                                                                                                                                       Patent
                                                                                                                                                                                                                                                                                                       English
             123177-61-7
                                                                                                                                                                                                                                        FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
             RL: PRP (Properties)
(total bond orders at sulfur in)
123177-61-7 CAPLUS
                                                                                                                                                                                                                                                                                                                                                        APPLICATION NO.
                                                                                                                                                                                                                                                    PATENT NO.
                                                                                                                                                                                                                                                                                                       KIND
                                                                                                                                                                                                                                                                                                                         DATE
              Methanetrisulfonic acid. ion(3-) (9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                                                       US 1988-164663
CA 1989-591956
WO 1989-US763
                                                                                                                                                                                                                                                                                                        A
A1
A1
                                                                                                                                                                                                                                                    US 4827021
CA 1326242
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                                                                                                                                                                                                                                                                                                                          19940118
                                                                                                                                                                                                                                                    WO 8908636
                                                                                                                                                                                                                                                                                                                         19890921
                503-
                                                                                                                                                                                                                                                   W: AU, JP, KR
RW: AT, BE, CH, DE, FR, GB, IT, LU, NL, SE
AU 8933504 Al 19891005 AU 1989-33504
 -035-CH-SO3-
                                                                                                                                                                                                                                                    AU 609288
                                                                                                                                                                                                                                                                                                                         19910426
             By combining linear relationships between log k and log r, bond order (n) and force consts. (k) where k is the CS stretching force constant of a
                                                                                                                                                                                                                                                                                                                                                       EP 1989-903301
                                                                                                                                                                                                                                                    EP 403528
AB
                                                                                                                                                                                                                                                    EP 403528
                                                                                                                                                                                                                                                                                                                         19930210
                                                                                                                                                                                                                                                   R: AT, BE, CH, DE, FR,
JP 03503282 T2
JP 2738967 B2
AT 85603 E
                                                                                                                                                                                                                                                                                                                         GB, IT, LI, LU, NL, SE
19910725 JP 1989-503132
             of length r, a correlation is established between CS bond lengths and simple valence-bond bond orders. For a variety of S(VI) species with bonds to electroneg. ligands, including Me groups, this is used to demonstrate that the total bond order at S in these species is six, consistent with a duodecet rule. Apparently, S-CH3 groups in compds.
                                                                                                                                                                                                                                                                                                                         19980408
                                                                                                                                                                                                                                                                                                                                                        AT 1989-903301
                                                                                                                                                                                                                                                                                                                          19930215
                                                                                                                                                                                                                                                     KR 9700138
                                                                                                                                                                                                                                                                                                          B1
                                                                                                                                                                                                                                                                                                                         19970104
                                                                                                                                                                                                                                                                                                                                                        KR 1989-72047
US 1988-164663
                                                                                                                                                                                                                                       PRIORITY APPLN. INFO.:
                                                                                                                                                                                                                                                                                                                                                                                                           A 19880307
 such
              as CH3SO2F, CH3SO2OH, Me2SO2, and related species compete with other ligands for the delocalization of electron pairs into the valence sh
                                                                                                                                                                                                                                                                                                                                                        EP 1989-903301
                                                                                                                                                                                                                                                                                                                                                                                                           A 19890227
of
the central sulfur atom, in this case through hyperconjugation. This
accounts for the enhanced acidities of hydrogen atoms bonded to
a-carbon atoms in such compds., and their ability to form hydrogen
bonds. Evidence for hydrogen bonding in CH3SO2F, CH3SO2C1, and
(CH3) 2502;
                                                                                                                                                                                                                                                                                                                                                        WO 1989-US763
                                                                                                                                                                                                                                                                                                                                                                                                           A 19890227
                                                                                                                                                                                                                                      OTHER SOURCE(S): MARPAT 111:153222
IT 54322-33-7 Methanetrisulfonic acid
RL: CAT (Catalyst use): USES (Uses)
(catalyst, for addition of alkoxymethane with ketene)
RN 54322-33-7 CAPLUS
             2502,
and in CH3S020H (via both S-OH and S-CH3), is discussed. The use of C-H
bond lengths and stretching force consts. as a possible diagnostic tool
                                                                                                                                                                                                                                                     Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
              select mols, capable of CH hydrogen bonding is also discussed.
                                                                                                                                                                                                                                                      503H
                                                                                                                                                                                                                                       HORS-CH-SORH
                                                                                                                                                                                                                                                  R10CH2CR3R4C02R2 (I; R1, R2 = C1-8 alkyl; R3, R4 = H, alkyl, aryl),
                                                                                                                                                                                                                                                     as solvents in the formation of coating compns., are prepared by
                                                                                                                                                                                                                                       addition of
                                                                                                                                                                                                                                                    RIOCHZOR2 with R3R4C:CO in the presence of CH2(SO3H)2, CH(SO3H)3, or
                                                                                                                                                                                                                                       mixture
                                                                                                                                                                                                                                                    ire thereof. A mixture of 23.0 g CH2(oMe)2, 0.071 CH2(SO3H)2, and 0.30 and CH2:CO was stirred at 25-40^\circ under N to give 24.9 g I (R1 = R2 = Me, R3 = R4 = H). Similarly prepared was I (R1 = R2 = Et, R3 = R4 = H).
                                                                                                                                                                                                                                       L23 ANSWER 11 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1985:184717 CAPLUS
DOCUMENT NUMBER: 102:184717 CAPLUS
102:184717 CA
          ANSWER 10 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
ESSION NUMBER: 1986:431749 CAPLUS
DMENT NUMBER: 105:31749
EE: The electrochemistry of a dimeric and two monomeric cis-trioxomolybdenum(VI) complexes containing cyclic triamine ligands in protic and aprotic media: model compounds for the active site in formate
 ACCESSION NUMBER:
DOCUMENT NUMBER:
TITLE:
                                                                Herrmann, Willy: Wieghardt, Karl
Ruhr-Univ., Bochum, D-4630, Fed. Rep. Ger.
Polyhedron (1986), 5(1-2), 513-20
CODEN: PLYMDE: ISSN: 0277-5387
Journal
 dehydrogenase
AUTHOR(S):
CORPORATE SOURCE:
SOURCE:
                                                                                                                                                                                                                                        DOCUMENT TYPE:
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
 DOCUMENT TYPE:
 DOCUMENT ....
LANGUAGE: Englis.

IT 54322-33-7
RL: PRP (Properties)
(electrochem. reduction of molybdenum trioxotriazacyclononane or
                                                                                                                                                                                                                                                    PATENT NO.
                                                                                                                                                                                                                                                                                                        KIND
                                                                                                                                                                                                                                                                                                                                                         APPLICATION NO.
                                                                                                                                                                                                                                                                                                                         DATE
                                                                                                                                                                                                                                        AU 538363
AU 8171460
PRIORITY APPLN. INFO.:
                                                                                                                                                                                                                                                                                                          B2
A1
                                                                                                                                                                                                                                                                                                                          19840809
19811217
                                                                                                                                                                                                                                                                                                                                                        AU 1981-71460
 molybdenum
trioxotrimethyltriazacyclononane nuclear and dinuclear complexes on
                                                                                                                                                                                                                                                                                                                                                        AU 1981-71460
                                                                                                                                                                                                                                                    54322-33-7
RL: CAT (Catalyst use); USES (Uses)
  (catalyst for alkoxylation of alcs. and phenols)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
              mercury in)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                SO3H
 нозя-сн- sозн
                                                                                                                                                                                                                                                      șo3H
                                                                                                                                                                                                                                        нозя-сн- возн
             The electrochem. of cis-Moo3L (I), (L=N,N',N''-trimethyl-1,4,7-triazacyclononane) or cis-Moo3L' (II), (L'=1,4,7-triazacyclononane) was in protic (CH3SO3H) and aprotic media (DMF) by cyclic voltammetry at the hanging Hg drop (HMD) electrode. I and II are quasi-reversibly reduced
                                                                                                                                                                                                                                                    Alcs. and phenols were alkoxylated with alkylene oxides in the presence
catalysts consisting of organic sulfonic acids or their salts. The
                                                                                                                                                                                                                                                    acids chosen have an acid strength greater then that of methanesulfonic acid; the cations are chosen from first row transition metals, Be, Mg,
                                                                                                                                                                                                                                                    B, Al, Sn, or Zr. Thus, BuOH was ethoxylated with ethylene oxide (6:1
                                                                                                                                                                                                                                                    ratio) at 160° in the presence of 2m methanesulfonate to give 93.88 ethylene glycol monobutyl ether and small amts. of diethylene glycol monobutyl ether (20.9:1 mol ratio). Other catalyst used were 2m benzenesulfonate or trifluoromethanesulfonate and Al methanedisulfonate.
```

L23 ANSWER 8 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1989:573247 CAPLUS 111:173247

The duodecet rule. Part 2. C-H hydrogen bonding by

DOCUMENT NUMBER:

CORPORATE SOURCE:

TITLE:

AUTHOR (S):

L23 ANSWER 9 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1989:553222 CAPLUS DOCUMENT NUMBER: 111:153222

Process for the preparation of alkyl

DATE

19880307

19890224 19890227

19890227

19890227

19890227

19890227

DATE

19800613

19800613

DOCUMENT NUMBER:

INVENTOR(S):

TITLE:

L23 ANSWER 12 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1983:523106 CAPLUS OCCUMENT NUMBER: 99:123106 NPhenylcarbamate ester oligom PATENT ASSIGNEE(S): Asahi Chemical Industry Co., L

N-phenylcarbamate ester oligomers Asahi Chemical Industry Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXKAF SOURCE:

Patent Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE JP 58085852 JP 62008430 PRIORITY APPLN. INFO.: 19830523 JP 1981-183739 19811118 A2 B4 19870223 JP 1981-183739 19811118

54322-33-7

DOCUMENT TYPE:

LANGUAGE:

State (Catalyst use): USES (Uses)
(catalysts, for oligomerization of phenylcarbamate esters with methylene donors)
54322-33-7 CAPLUS

Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H HO28-CH-SO2H

N-Phenylcarbamate esters are oligomerized with methylene donors in the presence of solid-supported polyaulfonic acids. Thus, 10g kieselguhr and 3.5g CH2(SO3H)2 (I) [503-40-2] in 10 mL H2O was dried and baked 6 h at 200° to give catalyst containing 14% I. A mixture of the above catalyst 3, Et N-phenylcarbamate (II) 8.25, and dimethoxymethane 1g in 40 mL aulfolane was heated 1 h in an autoclave at 140° to convert 56% II to oligomers [87093-19-4] containing 73% 1,1°-methylenebis(4-ethoxycarbonylaminobenzene) [10097-16-2] and 11% 1,3°-methylenebis(4-ethoxycarbonylaminobenzene) [70381-86-1].

L23 ANSWER 14 OF 30
ACCESSION NUMBER:
DOCUMENT NUMBER:
1981:174556 CAPLUS
94:174556
2-Hydroxyacetophenone via Fries rearrangement and related reactions. A comparative applied study
Hocking, Nartin B.
CORPORATE SOURCE:
DOURCE:
DOCUMENT TYPE:
LANGUAGE:

CAPLUS COPPRIGHT 2006 ACS on STN
1981:174556
CAPLUS
94:174556
CAPLUS
PARTATION

LANGUAGE: OTHER SOURCE(S):

English CASREACT 94:174556

X SOUNCE(S): CASCACT 94:17356
54322-33-7
RL: CAT (Catalyst use); USES (Uses)
(Catalysts, for Fries rearrangement of Ph acetate)
54322-33-7 CAPUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H нозя-сн- возн

PhoAc was prepared directly from PhoH and AcOH in the presence of H2SO4; thermal rearrangement of PhoAc was studied in the presence of methanesulfonic and -trisulfonic acids, p-McGH4SO3H, and sulfonated cation exchange resin. Catalysis by H2SO4, H2SO4.H2O, H3PO4, NaHSO4, and CaCl2 was also studied. The catalysts gave o- and p-HOCGH4Ac, with larger

amts. of the ortho derivative at higher temps. The rates, product

ratios, and yields of these processes were compared. AlCl3, ZnCl2, and BF3 all gave faster reactions than the proton acids, and higher ortho-para ratios,

especially
with BF3 at low temps. The results obtained with the proton acids and
other compds. and with the Lewis acids were compared. The object of t
studies was to provide a com. route to pyrocatechol.

L23 ANSWER 13 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN
ACCESSION NUMBER: 1983:504727 CAPLUS
DOCUMENT NUMBER: 99:104727
AUTHOR(S): 484001 Action acid derivatives
AUTHOR(S): 784001 Action Action Cop. Nothin, Kiev, USSR
SOURCE: 2004001 Action Action Cop. Nothin, Kiev, USSR
CODEN: 2004001 TYPE: 2004001 COP. 2

Russian CASREACT 99:104727

LANGUAGE: Russian
OTHER SOURCE(s): CASREACT 99:104727
IT 54322-33-7
R1: RCT (Reactant): RACT (Reactant or reagent)
(reaction of, with sulfur tetrafluoride)
RN 54322-33-7 CAPLUS

Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H

нозя-сн-sозн

CH(SO3H)3 and SF4 gave CH(SO2F)3, which with Br, iodine, or Me3SiCl gave the corresponding RC(SO2F)3. CH2(SO2F)2 and Ag2O gave AgCH(SO2F)2, which with Me1 or C6F5SCl gave, resp., MeCH(SO2F)2 and C6F5SCH(SO2F)2.
4-MeC6H4SO3F and trifluoromorpholinosulfur gave 4-MeC6H4SO2F.

L23 ANSWER 15 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1981:139146 CAPLUS
DOCUMENT NUMBER: 94:139146 Tries[fluorosulfonyl]methane, HC(SO2F)3
AUTHOR(S): Kloeter, Gerhard; Pritzkow, Hans; Seppelt, Konrad
Anorg, Chem. Inst., Univ. Heidelberg, Heidelberg, D-6900/1, Fed. Rep. Ger.
Angewandte Chemie (1980), 92(11), 954-5
CODEN: ANCEAD; ISSN: 0044-8249

DOCUMENT TYPE:

DOCUMENT TYPE: OGLING
LANGUAGE: German

1 54322-33-7P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
(Reactant or reagent)

(preparation and reaction of, with sulfur tetrafluoride)

RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H

нозя-сн- возн

75533-77-6P 75533-77-6P
RE: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (preparation and reaction of, with sulfuric acid) 75533-77-6 CAPLUS METABLES (PREPARATION AND PREPARATION AND PREPARATION AND PROPARATION AND PROPAGATION AND PROP

503H

HO25-CH-503H

●3/2 Ba

75533-76-5

73533-76-3
RE: RCT (Reactant); RACT (Reactant or reagent)
[reaction of, with barium ion]
75533-76-5 CAPLUS
Methanetrisulfonic acid, monopotassium salt (9CI) (CA INDEX NAME)

503H

HO35-CH- SO3H

•к.

Successive treatment of HC(SO3K)3 with Ba2+, H2SO4, and SF4 gave, via HC(SO3Ba/2)3 and HC(SO3H)3, HC(SO2F)3 (I), which is a monobasic acid with strength falling between those of HSO3F and HNO3. Treatment of I with bases gave the salts MC(SO2F)3 (M = K, Rb, Cs and Ag); crystal structures

L23 ANSWER 15 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) of the K and Rb salts were detd. Treatment of AgC(SO2F)3 with MeI or X2 gave, resp. MeC(SO2F)3 and XC(SO2F)3 (X = Cl, Br, I). FC(SO2F)3 was prepd. by fluorinating I with XeF2.

The crystal and mol. structure of K3[CH(SO3)3].H2O was determined by y diffraction as an aid in the assignment of the IR and Raman spectra. The structure was solved using direct methods and refined by full-matrix

squares to R 0.045 for 839 observed unique reflections. The S-C-S angles of

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L23 ANSWER 16 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1981:68557 CAPLUS DOCUMENT NUMBER: 941:68557
DOCUMENT NUMBER:
TITLE:
                                                                94:68557
New electrolytes for direct methane fuel cells
Brummer, S. B.: Foos, J.; McNardy, J.; McVeigh, J.;
Toland, D.: Turner, M.
EIC Corp., Newton, MA, USA
Report (1979), DOE/ET/11321-T1, 74 pp. Avail.: NTIS
From: Energy Res. Abstr. 1980, 5(10), Abstr. No.
AUTHOR (S):
CORPORATE SOURCE:
15977
 DOCUMENT TYPE:
                                                                 Report
English
 LANGUAGE:
            54322-33-7
RL: USES (Uses)
                     (electrolytes, for direct-methane fuel cells, development and testing
            of)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
              $03H
HO3S-CH-SO3H
            A program aimed at developing a fuel-cell electrolyte for the direct oxidation of CH4 and/or impure H fuels is described. Of interest are
            tribasic methanesulfonic acids CX2(SO3H)2 and CX(SO3H)3 where X is H, F, or Cl. Synthetic routes to CH2(SO3H)2 [503-40-2], CH(SO3H)3 [54322-33-7], CC1(2(SO3H)2 [76371-35-2], and CC1(SO3H)3 [76371-36-3] were identified and optimized. The diphenyl ester of CF2(SO3H)2 was prepared and various approaches to CF(SO3H)3 [76371-37-4] were investigated. In parallel with the synthetic program, apparatus was designed and fabricated for the testing of the electrolytes under fuel-cell conditions. A new FTF test cell was developed for testing small amts. of electrolyte. Electrodes with low Pt loading were losed
            loped for use in electrolyte evaluation. Optimum performance with H3PO4 was achieved using 1 mg Pt/cm2 and 1 mg TFE 30/cm2 deposited on TFE tape, supported on a Au-plated Ta screen, and sintered. Preliminary half-cell tests using CH2(SO3H)2 and CH(SO3H)3 indicated that these acids are insufficiently stable for use as fuel-cell electrolytes. However, tests using CC12(SO3H2) and CC1(SO3H)3 were encouraging, yielding H oxidation
             equal to or better than those using H3PO4. Stability tests were
by heating a sample of each acid at 130° for 30 days under N, O, and H. At the end of the test, each sample was analyzed for decomposition In
conducted
              position In
no case did IR anal. indicate significant decomposition and in the case
              chloroacids, only a trace amount of free Cl- was observed Conductivity
measurements
showed the aqueous acids to be of the same conductivity as aqueous
H3P04. The dihydrate
of CR12(S03H)2 was more conductive than CF3SO3H.H2O [49789-04-0] but less
conductive than 100% H3P04.
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L23 ANSWER 17 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) .apprx.113* indicate an expansion from tetrahedral stereochem. and the S-C bond lengths of .apprx.1.81 Å are longer than those for K2(CH2(SG3)2] (1.77 Å) and Ca(CH3SG3)2 (1.75 Å). The IR (4000 to 50 cm-1) and Raman spectra of K3(CH(SG3)3].H20 and K3(CD(SG3)3].D20 were detd. at 77* K and interpreted in accordance with the structure, and with the use of the spectra of anhyd. K3(CH(SG3)3) and of aq. solns. of the sol. Li salts. The SG3 groups show their characteristic group frequencies: vsym(C-S) 762, vasym(C-S) 820, 6sym(CS3) 170, and & asym(CS3) apprx.210 cm-1, resp. IR spectra of samples contg. the isotopically dil. HDO species confirm the presence of 2 types of H bond per H2O mol.

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L23 ANSWER 18 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1975:139193 CAPLUS
DOCUMENT NUMBER:
                                                       82:139193
                                                        Reactions of dipotassium diazomethanedisulfonate in
                                                       Reactions of dipotassium diazomethanedisulfonate in aqueous solution
Young, J. Michael
Dep. Pharmacol., Univ. Cambridge, Cambridge, UK
Journal of the Chemical Society, Perkin Transactions
1: Organic and Bio-Organic Chemistry (1972-1999)
(1974), (221, 2541-3
CODEN: JCPRB4; ISSN: 0300-922X
AUTHOR (5):
 CORPORATE SOURCE:
SOURCE:
DOCUMENT TYPE:
LANGUAGE:
OTHER SOURCE(S):
                                                       CASREACT 82:139193
          R SOURCE(S): CASREACT 82:139193
55110-91-19
RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)
55110-91-3 CAPLUS
Methanetrisulfonic acid, tripotassium salt (3CI, 9CI) (CA INDEX NAME)
```

SO3H HO35-CH- SO3H

Decomposition of (KO3S)2CN2 (I) in H2O at 4° gave (KO3S)3CH and (KO3S)2C:NNH2, in N KOH at 70° gave (KO3S)2C:NNKSO3K, and in aqueous piperidine at 70° gave (KO3S)2CH2. I with MeOH gave (KO3S)2CH0Me.

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ANSWER 20 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
SSION NUMBER: 1963:468923 CAPLUS
MENT NUMBER: 59:68923
INAL REFERENCE NO: 59:12707d-f
E: Alkylation of phenols
NTOR(S): McConnell, Wayne V.; Davis, Herman E.
FS. 2 DD. 2 DD. 2 DD. 2 DD. 4 DD.
ACCESSION NUMBER:
DOCUMENT NUMBER:
ORIGINAL REFERENCE NO.:
TITLE:
INVENTOR(S):
PATENT ASSIGNEE(S):
                                                                                                                                                                                                                                               2 pp.
Patent
Unavailable
    SOURCE:
DOCUMENT TYPE:
    LANGUAGE:
PATENT INFORMATION:
                                                PATENT NO.
                                                                                                                                                                                                                                                                                                                                                                                                                                        APPLICATION NO.
                                              US 3082258 19630319 US 1960-2855
54322-33-7, Methanetrisulfonic acid
(catalyst in alkylation of phenols)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                                                                                                                                        US 1960-28557
                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                         19600512
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SO3H

The preparation of 2,6-di-tert-butyl-4-methylphenol (I) from

AB ine preparation of 2,0-datetr-outyl-w-methylphenol (1) home 4-methylphenol (II) and isobutylene using hydrated methanedi- or trisulfonic acid catalysis was described. I was useful as an antioxidant and stabilizer for fats and oils. Thus, 112 g. isobutylene was bubbled through a flas containing 108 g. II and 1.1 g. methanedisulfonic acid dihydrate (III) in 100

OC . benzene. In the initial stages the temperature varied from 25-40° due to the cooling effect of isobutylene refluxing in a dry ice-actione cooled condenser. Thereafter the temperature was held at 40° for a tot reaction time of 6 hrs. The supernatant liquid was decanted from the catalyst. Unreacted II (68) and 2-tert-butyl-4-methylphenol (31% conversion) were extracted with aqueous NaOH. After removal of C6H6, I

obtained (63% conversion), m. 68-9* (50% aqueous MeOH). Under the same conditions, 5.5 g. III gave an 88% conversion to I. Only a 20% conversion to Tresulted from use of 1,2-ethanedisulfonic acid. Benzenedisulfonic acid caused polymerization of isobutylene. When p-toluenesulfonic acid or H2504 was used in concentration of 5% based on the weight of II the product had

uct had poorer color and odor. White, odorless I could also be prepared in 84

80% conversions, resp., using 2.2 g. III and no solvent or using 1% by weight methanetrisulfonic acid trihydrate.

L23 ANSWER 19 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1975:64946 CAPLUS
COCLMENT NUMBER: 22:64946
RIVENTOR(S): Solid catalysts for heterogeneous reactions
RODA, Peter
PATENT ASSIGNEE(S): HI (TAMI) Institute for Research and Development CODEN: GWXXBX
CODEN: GWXXBX

DOCUMENT TYPE: Patent

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 2401958	A1	19740718	DE 1974-2401958	19740116
US 3920582	А	19751118	US 1974-430804	19740104
GB 1446964	A	19760818	GB 1974-1839	19740115
JP 50046587	A2	19750425	JP 1974-7615	19740117
PRIORITY APPLN. INFO.:			IL 1973-41330 A	19730117

54322-33-7

RL: RCT (Reactant); RACT (Reactant or reagent) (catalyst preparation with, for heterogeneous catalysis or organic

reactions) RN 54322-33-7 CAPLUS

Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H

нозs-сн- возн

Catalysts for heterogeneously catalyzed reactions were prepared by impregnation of carriers with sulfonic acids. Thus, 50 g SiO2-Al2O3 pellets were treated for 30 min with 14 g benzene-1,3-disulfonic acid in H2O at 80°, dried for 6 hr at 150°, and calcined for 6 hr at 200° to give 60 g catalyst. A H2O-C2H4 mixture of mol. ratio 1:1 was passed over this catalyst at 195° to give a C2H4-C2H5OH conversion of 0.3-0.5 mole % without splitting off acid from this catalyst.

L23 ANSWER 21 OF 30 CAPLUS COPYRIGHT 2006 ACS on STA ACCESSION NUMBER: 1963:14557 CAPLUS DOCUMENT NUMBER: 58:14557
ORIGINAL REFERENCE NO.: 58:2371g-h
TITLE: Esterification catalysts
INVENTOR(S): Toucy, George P.; Goins, Rex H.
PATENT ASSIGNEE(S): Eastman Kodak Co.

PATENT ASSIGNEE (5): SOURCE: SOURCE: DOCUMENT TYPE: Unavailable

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE US 3053884 19620911 US 1959-8453: 54322-33-7, Methanetrisulfonic acid (as catalyst for esterification) 54322-33-7 CAPLUS Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME) 19591009 US 1959-845336

IT

SO3H

HO3S-CH-SO3H

CH2(SO3H)2 and CH(SO3H)3 are superior catalysts for preparing esters by treating saturated aliphatic mono- and polyhydroxy alcs. With phenyl dicarboxylic acids or saturated aliphatic carboxylic acids and their anhydrides. A lower concentration of catalyst is required and the ester used

used is nearly colorless and is heat stable. Two moles phthalic anhydride and five moles BuOH were refluxed 7 hrs. in the presence of various acid catalysts. The catalyst used, the catalyst concentration based on the

alic anhydride, and the percent phthalic acid in the product are: CH2(SO3H)2, 0.1, 0.02; CH(SO3H)3, 0.1, 0.03; H2SO4, 0.1, 0.35; MeSO3H, 0.2, 1.6; MeCGH4SO3H, 1.0, 2.0; (CH2SO3H)2, 0.2, 0.85. Data are given which show the superiority of these two catalyst for the esterification of n-octyl alc. with adipic acid and glycerol with 2-ethylhexanoic acid.

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CODEN: JACSAT; ISSN: 0002-7863

VERT TYPE: Journal
UAGE: Unavailable
54322-33-7, Methanetrisulfonic acid
(ionization in AcOH)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                            ŞO3H
 нозя-сн- sозн
 AB Values of Ho for dilute solns. (5 + 10-4 to 5 + 10-3 M) of 11 strong acids in AcOH containing 0.12% water were measured by use of indicators \frac{1}{2}
                        rators \alpha-naphtholbenzein (I) and o-nitroaniline. Ho = -log(BH+)/(B) + pKa, where (BH+) and (B) are the concus, of the acidic and basic forms of indicator, and pKa is the thermodynamic dissociation constant for the
   conjugate
acid of the indicator. The pKa for I was evaluated as 0.53. The order
                       increasing acid strength at equal molarities is: HCl, methanesulfonic, sulfuric, carboxymethanesulfonic, chloroamethanesulfonic, chloroamethanesulfonic, chloroamethanedisulfonic, methanedisulfonic, and methanetrisulfonic acids. Ho values for anhydrous solns. of 4 monobasic acids at 5 + 10-3 M were measured, and from the increased acidity found, equilibrium consts. for the reaction
                            acids with water were calculated H2SO4 was found to be monobasic.
acids with water were calculated H2SO4 was found to be monobasic. Dissociation consts., Kc, of HCl, HBr, HClO4, and H2SO4 in AcOH (calculated from conductivity data of Kolthoff and Willman (C.A. 28, 3644.1)) are 5.1 + 10-10, 1.9 + 10-7, 9 + 10-7, and 7.4 + 10-9, resp. The fact that values of ApKc from conductivity and from HO data are in reasonable agreement shows that equilibrium in AcOH involve, primarily, undissocd. species.
L23 ANSWER 24 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1949:27346 CAPLUS COPYRIGHT NUMBER: 43:27346 CAPLUS CAPLU
                                                                                                                    Reaction of acetylene and acetic acid. Societe des usines chimiques Rhone-Poulenc
   DOCUMENT TYPE:
  FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                         PATENT NO.
                                                                                                                     KIND DATE
19410926
                                                                                                                                                                                                   APPLICATION NO.
                       FR 867066 19410926 FR
54322-33-7, Methanetrisulfonic acid
(catalyst of HgO, HgSO4 and, in C2H2 reaction with AcOH)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                                                                                                                                                                                                                                                                                                                           19400224
                             SO3H
  нозя-сн- возн
                         With a mixture of sulfonic acids containing CH-(SO3H)3 36, HgSO4 21, and
                        14
as catalyst, AcOH 5600 g. and C2H2 give CH2CHOAc or MeCH(OAc)2. The sulfonic acid mixture is prepared by adding 63% H2SO4 1070 to Ac20 400 g. slowly so that the temperature does not rise above 115°, letting stand 3 hrs. at 120°, adding glacial AcOH 500 g. with the temperature at 120° another 3 hrs., and then more glacial AcOH 3000 g., with stirring 0.5 hr. The mixture contains 15% CH (SO3H)3.
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L23 ANSWER 22 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1953:70596 CAPLUS DOCUMENT NUMBER: 47:70596

47:11919f-i

47:11919f-i
Acid-base equilibria in glacial acetic acid
Smith, Thor L.; Elliott, John H.
Hercules Powder Co., Wilmington, DE
Journal of the American Chemical Society (1953), 75,
3566-71
CODEN: JACSAT; ISSN: 0002-7863

ORIGINAL REFERENCE NO. :

AUTHOR(S):

CORPORATE SOURCE:

DOCUMENT TYPE:

```
L23 ANSWER 23 OF 30 CAPLUS COPYRIGHT 2006 ACS ON STN ACCESSION NUMBER: 1950:28491 CAPLUS DOCUMENT NUMBER: 44:28491 ORIGINAL REFERENCE NO.: 44:5552b-d
DOCUMENT NUMBER:
ORIGINAL REFERENCE NO.:
                                                              Aluminum methionate
Christian, John E.; Jenkins, Glenn L.
Purdue Research Foundation
 TITLE:
 INVENTOR (S):
 PATENT ASSIGNEE(S):
DOCUMENT TYPE:
                                                              Patent
                                                              Unavailable
 LANGUAGE:
 FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                                                                                             APPLICATION NO.
              PATENT NO.
                                                              KIND DATE
                                                                                                                                                                       DATE
                                                                                                             US 1946-701091
                                                                               19500418
                                                                                                                                                                       19461004
              US 2504107
             US 230410/

#855840-41-4, Methanetrisulfonic acid, aluminum salt

[preparation of]

#855840-41-4 CAPLUS

Methanetrisulfonic acid, aluminum salt (5CI) (CA INDEX NAME)
ΤT
               ș03-
  -o3s-сн- so3-
         ● Al 3+
            Certain Al alkyl sulfonates are effective as antiperspirants, e.g. Al methionate (I), Al ethanedisulfonate, Al methanetrisulfonate, Al methanesulfonate, Al sulfoacetate, Al o-sulfobenzoate. General requirements are: the Al ion or its equivalent; one or more sulfo
             alkyl group. The more sulfo groups, the greater is the astringency. I
prepared by adding a solution of Al2(SO4)3 to a solution of Ca methionate in water, digesting the mixture for 10 min., and filtering it. The solution of I
             is concentrated, and EtOH is added until I crystallizes out. I is
 hygroscopic,
soluble in water, does not crystalline out of astringent creams. The
              salts are prepared by treating the respective Ba compds. with Al2(SO4)3 solution These antiperspirants are harmless to skin and fabrics.
ACCESSION NUMBER: 1947:11308 CAPLUS

DOCUMENT NUMBER: 1947:11308 CAPLUS

ORIGINAL REFERENCE NO.: 41:2303e-g

Diameter changes of gelatinized coacervate drops of the complex coacervate gelatin-gum arabic, resulting from a change in pN of, or from the addition of neutral salts to, the surrounding medium. I de Jong, H. G. Bungenberg; Landsmeer, J. M. F.

CORPORATE SOURCE: Univ., Leiden

ROUNCE: Recueil des Travaux Chimiques des Pays-Bas et de la Belgique (1946), 65, 606-13

CODEN: RTCPB4: ISSN: 0370-7539

DOCUMENT TYPE: Journal LANGUAGE: English

IT 55110-91-3, Methanetrisulfonic acid, tripotassium salt (effect on diameter changes of coacervate drops of gelatingum arabic)

RN 55110-91-3 CAPLUS

CN Methanetrisulfonic acid, tripotassium salt (3CI, 9CI) (CA INDEX NAME)
              SO3H
 HOSS-CH-SOSH
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Diameter changes of coacervate drops (100-200 μ) reached equilibrium ly (5-10 min.) in media of a given pH or salt concentration The changes followed microscopically at salt concns. low enough (10-40 milliequiv. 1.) to allow reversibility in the swelling and shrinking. The diameter min. at the pH (3.7) where the H2O content is a min. Reversible swelling due to addition of a salt (KCl, CaCl2, LaCl3, K2SO4, or K3CH(SO3)3) is for the 1-1 valence type and increases as the salt deviates from this valence type (double valence rule).

L23 ANSWER 26 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1944:16250 CAPLUS
ORIGINAL REFERENCE NO.: 38:2347f-g
TITLE: Reaction of Joeum with AcOH or Ac20
INVENTOR(S): Cockerille, Frank O.
PATENT ASSIGNEE(S): P.I. du Pont de Nemours & Co.

PATENT ASSIGNEE(S): DOCUMENT TYPE: LANGUAGE: Patent Unavailable

LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. APPLICATION NO. KIND DATE DATE 19431109 US 2333701 19431109 54322-33-7, Methanetrisulfonic acid US 1940-365947 19401116 IТ

(preparation of)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H HORS-CH-SORH

Methanetrisulfonic acid is prepared by a process involving adding HOAc or Ac2O to oleum, in a proportion of 4 to 7 parts of oleum to each part of the HOAc or Ac2O, the temperature being allowed to rise gradually to 65-85 during the addition of 50-751. and not above 90 during the addition of the remainder, of the Ac2O or HOAc, and completing the reaction by holding the mixture at 90-100.

ANSWER 27 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN (Continued) (0101:4111) = 54° 5:; (001):(111) = 57° 30'; (001):- (102) = 29° 25'; (010):(.hivin.110) = 43° 50'; plane of the optical axes (100); the sharp bissectrix is (010); $\rho > v$; the crystals are optically negative. The normal T1 salt is anhyd. Normal Ba salt (9H2O); at 25', 100 g, water dissolves 0.845 g, anhyd. salt; crystallographic properties: rhombic system; a:b:c = 0.834:1:0.564; observed forms (212), (010), (110), (111) and (120); angles: (010):(110) = 50° 11'; (010):(121) = 31° 30'; plane of the optical axes (010). Normal NH4 salt, anhyd.; crystallographic properties:rhombic system; oc

Normal Nns Sait, annyo., C., State Coll., (011), (101), (110) and (211); angles: (001): (011) = 55° 34'; (100): (101) = 25° 21'; (100): (110) = 43° 44'; (100): (211) = 25° 16'. PC15 with the free acid does not give the trisultonyl chloride, SO2 being evolved; therefore, derivs. could not be prepd.

L23 ANSWER 27 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1932:23284 CAPLUS

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.:

1932:23284 CAPUS 26:22324 26:2413g-1,2414a-d The chlorination of methanetrisulfonic acid Backer, H. J. Recueil des Travaux Chimiques des Pays-Bas et de la Belgique (1932), 51, 374-80 CODEN: RTCPB4: ISSN: 0370-7539 AUTHOR (S): SOURCE:

DOCUMENT TYPE: Journal Unavailable

LANGUAGE: Unavailable IT 54322-33-7, Methanetrisulfonic acid

(chlorination of)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H

HO35-CH-SO3H

cf. C. A. 25, 75. The chlorination of CH(SO3H)3 may be carried out by heating 7.8 g. of the tri-K salt and 0.1 g. I in 60 cc. water containing

Cl in a sealed tube at 200° for 10 hrs. After evaporating the contents of the tube and recrystg. the residue 6 times from its weight of water,

chloromethanetrisulfonate was obtained in a pure state in 708 yield; it does not contain H2O; at 25° 100 g. water dissolves 3.38 g.; crystallographic properties: monoclinic system; a:b:c = 0.8040:1:0.7659; B = 87° 32'; observed forms (.hvin.11), (010), (111), (101), and (011); angles: (010):(011) 52° 34'; (010):(111) = 61° 27'; (100):(.hivin.101) = 42° 36'; (100):(01) = 44° 33'; (010):(.hivin.111) = 60° 38'. For the preparation of the free acid and other salts the tristrychnine salt, which crystallizes without H2O, was used. The free acid (4.5 H2O) m. 160-5° but 180-2° in the presence of P2O5; at 25°, 100 g. water dissolves 254.7 g. of the anhydrous acid. Normal Li salt (8 H2O): 100 g. water dissolves 109.9 g.

the anhydrous salt; crystalline properties: rhombic system; a:b:c = 0.933:1:0.708;

0.933:1:0.708;
observed forms (110), (011) and (010); angles: (110):(010) = 46°
59'; (1.hivin.11):(100) = 52° 43'; (010):(011) = 54° 50';
(110): (011) = 66° 51'. The normal Na salt does not contain H20 on crystallization from a saturated hot solution but 3 H20 on evaporating a solution at room temperature;
at 25°, 100 g. water dissolves 21.8 g. of the trihydrate. Normal Rb salt (1H20): 100 g. water dissolves at 25°, 3.34 g. of the anhydrous salt; crystalline properties: rhombic system; a:b:c = 0.9827:1:0.5437;
observed forms (1011. (010) (133) (221)

L23 ANSWER 28 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1932:8729 CAPLUS DOCUMENT NUMBER: 26:8729 ORIGINAL REFERENCE NO.: 26:962g-i

Zo:90xg-1 The salts of methanetrisulfonic acid Backer, H. J.; Terpstra, P. Recueil des Travaux Chimiques des Pays-Bas et de la Belgique (1931), 50, 1069-77 CODEN: RTCPB8: 158N: 0370-7539 AUTHOR (S): SOURCE :

CODEN: RTCPB4; ISSN: 0370-7539
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable
IT 54322-33-7, Methanetrisulfonic acid
(and salts)
RN 54322-33-7 CAPLUS
CN Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)

SO3H

HO35-CH-503H

The normal K, Ag and Tl salts of methane-trisulfonic acid have been described (cf. Backer, C. A. 25, 915); the present paper deals with other salts, which were prepared in the usual way. Methanetrisulfonic acid contains 3H2O, m. 162-162.5°; 100 g. of water at 25° dissolves 221 g. of the anhydrous acid and 510 g. of the crystallized

Detailed crystallographic data are given for the normal NH4 salt

cains
0.5 H2O; crystals are of the monoclinic system), the normal Li salt (4
H2O; monoclinic system), normal Na salt (3 H2O; very thin monoclinic
plates), normal Rb salt (1 H2O; rhombic system, bisphenoidal class),
normal Cs salt (1 H2O; monoclinic system), normal Ag salt (1 H2O;
trielinie system), normal Ca salt (12 H2O; rhombic system), normal Ba

(9 H2O; monoclinic crystals), normal La salt (6 H2O; monoclinic

and normal cinchonine salt (8 H2O; rhombic, pseudotetragonal crystals). The mono-K dibrucine salt was obtained from the tri-K salt and an

amount of brucine acetate; it contains 9 H2O. The K and Rb salts are isomorphous and give mixed crystals. Sometimes small anhydrous triclinic crystals of the Cs salt were obtained and a triclinic dihydrate was prepared

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L23 ANSWER 29 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN ACCESSION NUMBER: 1931:8598 CAPLUS DOCUMENT NUMBER: 25:8598 ORIGINAL REFERENCE NO.: 25:915a-i
                                                                                                                     23.93-14
Methanetrisulfonic acid
Backer, H. J.; Klaassens, K. H.
Recueil des Travaux Kchimiques des Pays-Bas et de la
Belgique (1931), 49, 1107-17
CODEN: RTCP84; 158N: 0370-7539
 AUTHOR (S):
SOURCE:
  DOCUMENT TYPE:
                                                                                                                       Journal
                                                                                                                       Unavailable
                  54322-33-7, Methanetrisulfonic acid
                       (and salts)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                           SO3H
 HO3S-CH-SO3H
                        CH(SO3H)3 has already been prepared: (1) by Theilkuhl (Ann. 147, 134
                           on heating Ca Me sulfate with H2S2O7; (2) by Bagnall (J. Chemical
on heating Ca Me Suitate With Industry, ..., ..., 278(1899)) on sulfonating Ac derivs, of aromatic amines; (3) by oxidation of HSG(503H)3 (Albrecht, Ann. 161, 139(1872); of. Backer, C. A. 24, 76); (4) by the action of KZSO3 on O2NCH(SO3H)2 (Rathke, Ann. 167, 219(1873)), which is formed by the action of KZSO3 of O2NCCI3; it is, however, to be recommended to isolate the O2NCH(SO3H)2, which is formed as an intermediate product; (5) on heating NZC(SO3H)2 with acids, water or KHSO3
KHSO3 (von Pechmann, Ber. 28, 2382(1895); Fanti and Fisch, C. A. 24, 1841); (6) by sulfonation of CH2(503H)2. All these methods, except 1, are discussed in the present paper. Method 6: On heating CH2(503H)2 with SO3 (2.5 mols.) during 5 hrs. at 170°, CH(503H)3 may be obtained as the K selt with 10% yield. Method 2: sulfonation of PhNHAC 0.5 mol. AcNNFh is introduced slowly into 600 g. fuming H2504 containing 35% SO3, the mixture being heated 3 hrs. at 130° with mech. stirring; yield of the tri-K selt 57%; on carrying out the same reaction with succinanilide, a small yield of the same compound was obtained. Method 3: 0.1 mol. HSC(SO3K)3, in 1
                        water is oxidized at room temperature by a current of Cl with the
  gradual addition
gradual addition
of 80 g. KHCO3; yield 90%. Method 4: O2NCH(SO3H)2 was prepared
according to
Rathke (loc. cit.) by adding 82 g. O2NCCl3, gradually to 450 g. K2SO3 in
900 cc. water, heated to 75°, and keeping the temperature at 80°;
yield 40% of the anhydrous di-K salt, which dissolves in water at 25°
to the extent of 1.16%. The normal strychnine salt of O2NCH(SO3H)2
crystallizes with 3.5 H2O, the free acid itself with 2 H2O while the
normal Na and T1 salts crystallize without H2O; both the latter salts are
easily soluble in water. On heating the K salt with K2SO3 in a sealed
tube
tube
at 140°, CH(SO3K)3 is obtained in 67% yield. Method 5: N2C(SO3K)2,
prepared according to von Pechmann (loc. cit.), may be converted into CH
(SO3H)3 by the action of KHSO3 at 60° and finally on the water bath
(yield 62%), decomposing the compound with water or introducing it into
L23 ANSWER 30 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN
ACCESSION NUMBER: 1930:16847 CAPLUS
DOCUMENT NUMBER: 24:16847
TITLE: Methanoltrisulfonic acid
Fantl, Paul; Fisch, Julius
SOURCE: Journal fuer Praktisch Chemie (Leipzig) (1930), 124, 159-62
CODEN: JOURNAL TYPE: JOURN
                             503H
  нозя-сн- возн
                     ●3 K
                       54322-33-7, Methanetrisulfonic acid
                        (salts)
54322-33-7 CAPLUS
Methanetrisulfonic acid (7CI, 9CI) (CA INDEX NAME)
                             SO3H
  нозя-сн-возн
                           N2C(SO3K)2 and KHSO3 give CH(SO3K)3 and not HOC(SO3K)3 (cf. v. Pechmann, Ber. 28, 2374(1898)). The K, Ba and Ag salts were prepared and analyzed.
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L23 ANSWER 29 OF 30 CAPLUS COPYRIGHT 2006 ACS on STN (Continued)
HC1, the latter reaction giving a 400 yield. On introducing the diazo compd. into concol. HCl at 0°, CLCH(SO3H)2 is obtained. The following salts of CH(SO3H)3 and the acid itself were prepd. in the usual way: The free acid with 3.5 H2O, m. 156°; the tri-K salt, rhombic prisms with 1 H2O; 100g. of water dissolve at 25°,1.202 g. of the anhyd. salt and 1.261 g. of the crystd. salt. Crystallographic properties

(P. TERPSTRA): rhombic system, bisphenoidic class D2; a: b: c = 0.9977:11.13604. Forms: a = (100): b = (010): c = (001): p = (101): m = (110): o = (111): 0 = (1.hivin.11): q = (011): s = (121): angles: a: m = 44° 56'; c: p = 52° 38'; a: * = 51° 28'; b: e = 51° 34'; c: q = 52° 36'; o: s = 15° 22'.

The crystals obtained by evapn. at ordinary temp. have a larger dimension in the direction of the b-axis. The crystals show a marked piezoelectricity: the plane of the optical axes is (001): the a-axis is the sharp bissectrix; pcv. For Na light 2c 83° 1'; c = 1.513: β = 1.5253; γ = 1.5270: 2v = 51° 45'.

The tri-T1 salt also contains 1H2O; crystallographic properties (P. TERPSTRA): rhombic system, bisphenoidic class; a: b: c: = 0.9971:1:-1.3009: forms: m = (110): p = (101): q = (011): a = (1.hivin.11): * = (111): the crystals possess either the form or the form ', which have never been found to occur in the same crystal. Angles: m: m = 90° 10'; p:p = 105°4'; q:q = 123°0'; p:s = 38°20': q:q = 38°29'. The plane of the optical axes is (001): the sharp bissectrix (010). By means of a prism formed by (1.hivin.11) and (1.hivin.11) the refractive index β has been detd. to be 1.739, 1.743, 1.768 for x = 578, 546, 436, reap. The crystals are piezoelectrical. The tri-Ag salt contains 1 H2O; the Ba salt, 9 H2O: the latter is difficulty sol. in water, only to the extent of 0.1%. On mixing without precautions the solms. of the K salt and BaCl2, a double salt of Ba and K is obtained, Cho951KBa.3H2O, which,

er several recrystns. from a large amount of water, gives the pure Ba salt. The Ca salt contains 12 H2O and the La salt 6 H2O (cf. Backer and Klaassens, C. A. 24, 4729).

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chain nodes : 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 ring nodes : 1 2 3 4 5 6 7 8 9 10 chain bonds : 1-15 1-28 7-11 8-12 9-14 10-13 15-16 16-17 17-18 18-19 18-31 19-20 20-21 21-22 22-23 22-30 23-24 24-25 25-26 26-27 26-29 ring bonds : 1-6 1-2 2-3 3-4 4-5 4-10 5-6 5-7 7-8 8-9 9-10 exact/norm bonds : 1-6 1-2 2-3 3-4 5-6 9-14 exact bonds :

1-15 1-28 7-11 8-12 10-13 15-16 16-17 17-18 18-19 18-31 19-20 20-21 21-22 22-23 22-30 23-24 24-25 25-26 26-27 26-29 normalized bonds :

4-5 4-10 5-7 7-8 8-9 9-10

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L25 446 SEA SSS FUL L24

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L26 17222 L25

=> s 126 and 121

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BATCH **COMPLETE**

PROJECTED ITERATIONS: 6 TO 266
PROJECTED ANSWERS: 0 TO 0

L27 0 SEA SSS SAM L21

L28 0 L27

L29 0 L26 AND L28

=> s 121 and 119

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SEARCH TIME: 00.00.01

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BATCH **COMPLETE**

PROJECTED ITERATIONS: 6 TO 266
PROJECTED ANSWERS: 0 TO 0

L30 0 SEA SSS SAM L21

L31 0 L30

37 L19

L32 0 L31 AND L19

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